

# Nano-sized calcium phosphates (CaPs) coatings on the anodic oxidized titania porous surface via a microwave irradiation process

Cheng-Yu Chen<sup>a</sup>, Lih-Jyh Fuh<sup>a</sup>, Heng-Li Huang<sup>a</sup>, Jui-Ting Hsu<sup>a</sup>, Chiing-Chang Chen<sup>b</sup> and Dan-Jae Lin<sup>\*c</sup>

<sup>a</sup> School of Dentistry, College of Medicine, China Medical University, Taichung, Taiwan

<sup>b</sup> Department of Science Application and Dissemination, National Taichung University of Education, Taichung, Taiwan

<sup>c</sup> Department of Dental Hygiene, China Medical University, Taichung, Taiwan, Fax: +886-4-22073556; Tel: +886-4-22053366 ext 7706;

E-mail: [djlin@mail.cmu.edu.tw](mailto:djlin@mail.cmu.edu.tw)

## Introduction

Nano-sized calcium phosphates (CaPs) coatings on biomaterials are gaining attentions as nano-structured surfaces enhance the cellular activity more than surface along with micron-sized structure. The hypothesis of this research is that nano-sized CaPs can be precipitated on the anodic oxidized titania porous surface via a microwave irradiation (MI) process.

## Materials & Methods

The titania porous surface was prepared by anodic oxidation (280V, 10mA/cm<sup>2</sup>) of grade 2 commercial pure titanium discs in electrolytes composed of 0.2M calcium acetate and 0.04M  $\beta$ -glycerol phosphate disodium for 3mins. The titanium discs were microwave irradiated in water (MI-DD) and diluted Ca/P solutions (MI-CP) by a domestic microwave oven (2.45GHz, 1000W, 50%) for 5 and 20 minutes. The morphology, structure, and chemical characters of samples were evaluated by SEM, XRD and FTIR, the results were compared to hydrothermal (HT, 250°C, 3hours) and control (AO, anodic oxidize only) groups.

## Results & Discussion

By SEM, numberless nano-sized CaPs precipitates on titania porous surface were obtained on MI-CP (20mins) sample, whereas no precipitate was found on MI-DD samples (Figure 1 (a) and (b)). The nano-sized precipitates are shuttle-like with long axis around 100nm as shown in Figure 1 (c). There are various micron-sized hexagonal rod-like precipitates were found by HT (Figure 1 (d)). The XRD (Figure 2) shows weak hydroxyapatite (HA) peaks at all MI samples (20mins), and FTIR (Figure 3) shows broaden Ti-O, PO<sub>3</sub><sup>-4</sup>, and OH<sup>-</sup> absorption peaks indicates the Ca and P which originally incorporated in the anatase titania were leach-out to precipitate CaPs homogeneously on the porous surface by MI. The XRD of HT samples shows 25.88°, 31.72°, 32.19°, 32.96° of HA peaks beside anatase and titanium peaks. The FTIR of HT shows sharp absorption peaks of PO<sub>3</sub><sup>-4</sup> at 566, 604, 1050, 1092 cm<sup>-1</sup> also confirmed that HA obtained by HT can possess highly crystalline.

## Conclusion

Nano-sized CaPs precipitates on titania porous surface were obtained via a microwave irradiation (MI) process. Hopefully the nano-sized CaP precipitates firmly formed on the micron-sized porous titania would benefit to the protein

adsorption and osteoblasts differentiation and shortening the osseointegration time.

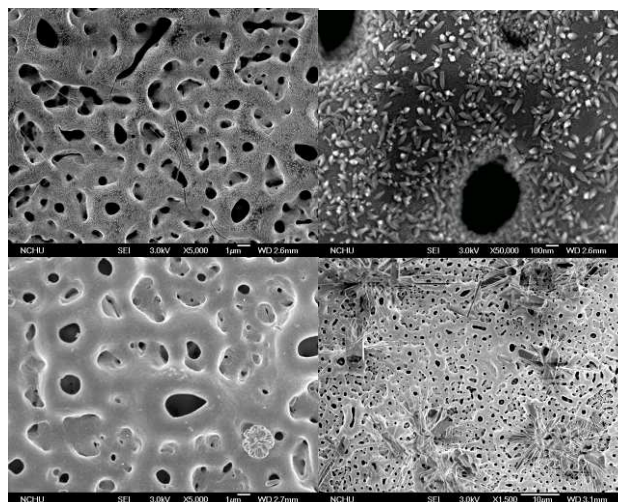


Fig. 1. SEM pictures of (a)MI-CP 20mins, (b)enlarge picture of (a), (c)MI-DD 20mins, and (d)hydrothermal at 250°C for 3hours.

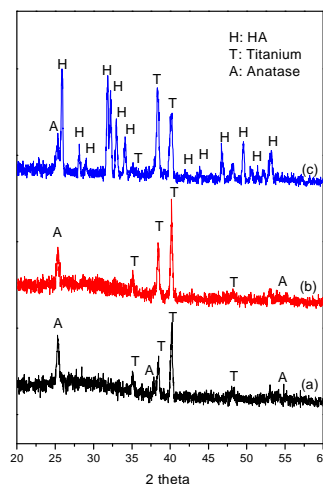


Fig. 2. XRD patterns of samples (a)MI-DD 20mins, (b)MI-CP 20mins, and (c)hydrothermal at 250°C for 3hours.

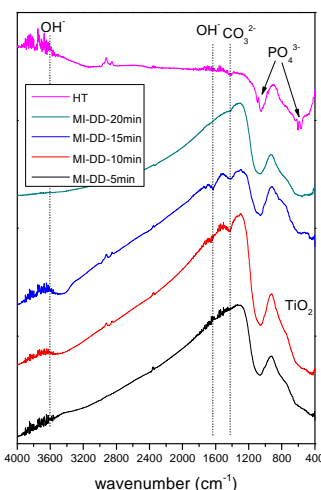


Fig. 3. FTIR spectrums of MI-DD 5, 10, 15, 20mins with compare to HT sample at 250°C for 3hours.

## References

- 1 H. Ishizawa and M. Ogino, *J Biomed Mater Res*, 1995, **29**, 1071–9.
- 2 D.J. Lin, M.T. Tsai, T.M. Shieh, H.L. Huang, J.T. Hsu, Y.C. Ko, and L.J. Fuh, *Journal of Biomaterials Applications* (in press).