

## **Abstract**

 This study first assessed workers' inhalatory and dermal exposures to polycyclic aromatic hydrocarbon (PAH) contained in oil mists. Then, their resultant lung cancer and skin cancer risks were estimated. Finally, control strategies were initiated from the health risk management aspect. All threading workers in a fastener manufacturing plant were included. 16 inhalatory and 88 dermal PAH exposure samples were collected. Results show that the inhalatory gas phase total PAH exposure level  $(8.60\times10^4 \text{ ng/m}^3)$  was much higher than that of particle phase  $(2.90\times10^3 \text{ m})$  8 ng/m<sup>3</sup>). Workers' mean inhalatory exposure level  $(8.83 \times 10^4 \text{ ng/m}^3)$  was lower, but its 9 corresponding 1-sided upper 95% confidence level (UCL<sub>1,95%</sub>=1.02×10<sup>5</sup> ng/m<sup>3</sup>) was higher than the time-weighted average permissible exposure level (PEL-TWA) regulated in Taiwan for PAHs  $(1.00 \times 10^5 \text{ ng/m}^3)$ . The mean whole body total PAHs dermal exposure levels was 5.44 $\times 10^6$  ng/day and the top five exposed surface areas were lower arm, hand, upper arm, neck, and head/front. The 13 estimated lifetime skin cancer risk  $(9.72 \times 10^{-3})$  was lower than that of lung cancer risk  $(1.64 \times 10^{-2})$ , 14 but both were higher than the significant risk level  $(10^{-3})$  defined by the US Supreme Court in 15 1980. The installation of a local exhaust ventilation system at the threading machine should be considered as the first priority measurement because both lung and skin cancer risks can be reduced simultaneously. If the personal protection equipment would be adopted in the future, both respiratory protection equipment and protective clothing should be used simultaneously. *Keywords:* Polycyclic aromatic hydrocarbons (PAHs); Oil mists; Exposure assessment; Health risk assessment; Fastener manufacturing industry

## **1. Introduction**

2 Based on the Taiwan governmental statistics in 2002, there were ~1,270 fastener manufacturing plants and in total employed with ~37,000 employees in the whole country. The 4 total fastener production rates increased from ~451,000 tons/yr in 1991 to ~1,269,000 tons/yr in 5 2003 accounting for  $\sim$ 14% world production. The manufacture of fasteners involves seven important industrial processes, including the wire drawing, forming, threading, cleaning, heat treatment, surface treatment, and packaging/shipping. Among them, mineral oil-based metalworking fluids (MWFs) are used in forming, threading, and heat treatment processes for cooling, lubricating, and corrosion inhibition purposes and hence might result in the emission of oil mist to the workplace atmosphere and the exposures of workers (Thornburg, 2000; Michalek et al., 2003). Considering mineral oils are produced by petroleum distillation processes, semi-volatile organic compounds, such as polycyclic aromatic hydrocarbons (PAHs), are expected to be contained in emitted oil mists. Epidemiological and animal studies have indicated that oil mist exposures might result in the laryngeal cancer (Russi et al., 1997), asthma (Robertson et al., 1988), bronchial hyper-responsiveness (Kennedy et al., 1999), lipoid pneumonia (Cullen et al., 1981), lung cancer (Kazerouni et al., 2000), and many other respiratory illnesses (Jarvholm et al., 1982; Massin et al., 1996). Particularly, workers exposed to high airborne concentrations of PAHs might result in excessive lung cancer rates (Verma et al., 1992; Moulin et al., 1993). In addition, intensive studies have also reported that all mineral oils are mutagenic to skin because of their 20 inherent PAH contents (Bingham et al., 1969; Roy et al., 1988; Granella et al., 1995). However, it should be noted that only very limited studies have been conducted to address workers' dermal exposures to PAHs (Jongeneelen et al., 1988; Tsai et al., 2001a). In fastener manufacturing industries, workers are exposed to PAHs contained in oil mists via both inhalatory and dermal exposure routes. Therefore, the present study was set out first to assess workers' inhalatory and dermal PAH exposures. Then, their resultant lung cancer and skin cancer risks were estimated. Finally, control strategies were initiated from the health risk management

aspect.

#### **2. Material and Methods**

## **2.1 Personal inhalable and dermal PAH exposure samplings**

 For fastener manufacturing industries, the forming, threading, and heat treatment processes are involved in the use of mineral oil-based MWFs for cooling, lubricating, and corrosion inhibition purposes. Among them, the threading workers were found with the highest oil mist exposures in our previous study (Chen et al., 2007), and hence were selected for illustration. All threading workers were included to conduct personal samplings for assessing their inhalable and dermal PAH exposures.

 The sampling method modified from the NIOSH method 5515 was used to assess workers' inhalatory PAH exposures. The sampling train consisted of a filter cassette (IOM personal inhalable aerosol sampler, SKC Inc., Catalog No. 225-70) and followed by a sorbent tube (polyurethane form (PUF) plug/3.5 g ??? (XAD-2) resins/PUF separation layer/0.5 g XAD-2 resin/ PUF plug). The sampling flow rate was specified at 2.0 L/min. Before sampling, all filters and sorbent tubes were cleaned and extracted with a solvent solution (mixture of n-hexane and 16 dichloromethane, v:v = 1:1) for 24 hrs in a Soxhlet extractor. Oil mists collected by the IOM 17 personal sampler were determined by using an electrical balance (Sartorius MP 8-6,  $\pm 0.01$ mg), then sent to the laboratory, together with the sorbent tube, for PAHs analysis to determine the concentrations of both particle phae PAHs and gas phase PAHs, respectively. A total of 16 personal inhalable samples (gas+particle phase PAH samples) were collected from this study. The surrogate skin sampling technique was adopted in this study for dermal exposure assessment. The method was first recognized as a standard method for assessing dermal pesticide exposures (WHO, 1986; US EPA, 1987), but its use has been extended to assess occupational exposures to PAHs (Jongeneelen et al., 1988) and dichlorobenzidene (London et al., 1989). A soft 25 polypropylene with a surface area 100 cm<sup>2</sup> was used as the collecting medium (i.e., exposure pad). Table 1 shows the 11 sampling body sites and their corresponding surface areas (based on a

1 standard man: height = 173 cm, body weight = 70 kg) for conducting dermal exposure assessment for each selected worker as recommended by US EPA (US EPA, 1987). However, subjected to worker's willingness, 11 sampling spots closed to the above mentioned sampling body sites were chosen for each selected worker (also shown in Table 1). As a result, 88 pad samples were obtained from this study. Before sampling, all pads were cleaned and extracted with a solvent 6 solution (mixture of n-hexane and dichloromethane,  $v.v = 1:1$ ) for 24 hrs in a Soxhlet extractor. Immediately after sampling, the pads were removed, packed in an aluminum foil, and stored at  $8 \times 4^{\circ}$ C until analysis.

# **2.2 Sample analyses**

 For PAHs analyses, all collected samples (including the filter, sorbent tube, and exposure pad) 11 were placed in a solvent solution (the mixture of *n*-hexane and dichloromethane,  $v: v = 1:1$ , respectively), and extracted in a Soxhlet extractor for 24 hrs. The extract was then concentrated, cleaned-up and re-concentrated to exactly 1.0mL or 0.5mL. PAH content was determined by using a gas chromatograph (GC) (Hewlett-Packard 5890A) with a mass selective detector (MSD) (Hewlett-Packard 5972) and a computer workstation. This GC/MS was equipped with a 16 Hewlett-Packard capillary column (HP Ultra 2 - 50 m×0.32 mm×0.17 um), an HP-7673A 17 automatic sampler, injection volume 1  $\mu$ L, splitless injection at 310 °C, ion sources temperature at 18 310 °C, oven temperature from 50 °C to 100 °C at 20 °C/min; 100 °C to 290 °C at 3 °C/min; and 19 hold at 290 °C for 40 min. The masses of primary and secondary ions of PAHs were determined using the scan mode for pure PAH standards. Qualification of PAHs was performed using the selected ion monitoring (SIM) mode (Jongeneelen et al., 1988; Tsai et al., 2001a; Tsai et al., 2001b; Tsai et al., 2002; Tsai et al., 2004; Li et al., 2003; Lee et al., 2004; Lin et al., 2006; Chen et al., 2006). The concentrations of 22 PAH compounds were determined, including naphthalene (NaP),

acenaphthylene (AcPy), acenaphthene (AcP), fluorene (Flu), phenanthrene (PA), anthracene (Ant),

fluoranthene (FL), pyrene (Pyr), cyclopenta[c,d]pyrene (CYC), benz[a]anthracene (BaA),



## **2.3 Data analysis**

# **2.3.1 Calculating PAH homologue distributions**

 In this study, the concentration of total PAHs was defined as the sum of the concentrations of the selected 22 PAH compounds. In addition, PAH contents were further sorted into three categories according to their molecular weights, including the low molecular weight (LMW-PAHs; containing two- and three-ringed PAHs), middle molecular weight (MMW-PAHs; containing four-ringed PAHs), and high molecular weight (HMW-PAHs; containing five-, six- and seven-ringed PAHs).

# **2.3.2 Calculating BaP equivalent concentrations**

Because BaP has been known to be the most carcinogenic PAH compound, the carcinogenic

 potency of each collected sample was also determined in terms of its BaP equivalent 2 concentration  $(BaP_{eq})$ . The carcinogenic potency of the total PAH exposure could then be 3 estimated as the sum of each individual  $BaP_{eq}$ . To calculate the  $BaP_{eq}$  for each individual PAH species, it requires the use of its toxic equivalent factor (TEF) for the given species relative to BaP carcinogenic potency. To date, only a few proposals for TEFs are available (Chu et al., 1984; Thorslund et al., 1991; Nisbet et al., 1992). From these, the list of TEFs completed by Nisbet and LaGoy (Chu et al., 1984) were adopted in this study (Table 2), as these have been demonstrated to be a better reflection of the actual state of knowledge on the toxic potency of each individual PAH species relative to BaP (Petry et al., 1996).

# 10 **2.3.3 Calculating personal inhalable and dermal PAH exposure concentrations**

 In this study, the total inhalatory PAH exposure was defined as the summation of the gas phase and particle phase PAH exposures. To estimate a worker's total dermal exposure level, the unit dermal surface area PAH exposure level for a given PAH species *i* on the body surface area *j* 14 ( $\text{UDC}_{ij}$ ; ng/100cm<sup>2</sup>/day) was directly measured from this study. The unit dermal surface area PAH exposure level for total PAHs (i.e., the sum of 22 PAH species) for a given body surface  $\arctan j$  (UDE<sub>total,j</sub>; ng/100cm<sup>2</sup>/day) can be determined by using equation 1:

17 
$$
UDE_{\text{total},j} = \sum_{i=1}^{22} UDC_{ij}
$$
 (1)

18 Dermal PAHs exposure levels for total PAHs at the body surface area *j* (*DEtotal,,j*; ng/day) can be 19 determined by equation 2:

20 
$$
DE_{total,j} = \sum_{i=1}^{22} UDC_{ij}AD_j
$$
 (2)

21 Where,

- AD<sub>j</sub>=the corresponding body surface area for the body surface area *j* (unit: 100cm<sup>2</sup>) of a 23 standard man (Table 1).
- 24 Dermal PAH exposure levels for species *i* on the whole body surface area (*DEi,whole*; ng/day) was 25 determined by using equation 3:

1 
$$
DE_{i,whole} = \sum_{j=1}^{11} \text{UDC}_{ij} \text{AD}_{j}
$$
 (3)

2 Total dermal PAHs exposure levels on the whole body surface area (DE*total,whole*; ng/day) was

3 determined by using equation 4:

4 
$$
DE_{total, whole} = \sum_{i=1}^{22} \sum_{j=1}^{11} \text{UDC}_{ij} \text{AD}_{j}
$$
 (4)

## 5 **2.3.4 Estimating personal inhalable and dermal PAH exposure profiles**

 In this study, the log-normality of the exposure profile for each exposure group was examined by using the W-test (Gilbert, 1987). The arithmetic mean was used to describe the average concentration (Rappaport, 1991). The method of the minimum variance unbiased 9 estimate (MVUE) was adopted to estimate the arithmetic mean  $(AM_{MVUE})$  and its lower and upper 10 one-sided 95% confidence intervals (LCL<sub>1,95%</sub> and UCL<sub>1,95%</sub>, respectively). Full calculating procedures were described in the study conducted by Attfield and Hewet(1992). The above method has also been recommended by the American Industrial Hygiene Association (AIHA) Exposure Assessment Strategies Committee for exposure data with various sample sizes and geometric standard deviations (GSDs) (Mulhausen and Damiano, 1998).

# 15 **2.4 Heath-risk assessment**

 Regarding the lung cancer risk via the inhalation route, WHO suggested the unit risk of 17 8.7×10<sup>-2</sup> ( $\mu$ g/m<sup>3</sup>)<sup>-1</sup> for the lifetime (70 years) PAHs exposure, assuming one exposed to one unit 18 BaP concentration (i.e.,  $1 \mu g/m^3$ ) (WHO, 1987). It worth noting above unit risk was proposed to estimate the lung cancer risk caused by the lifetime exposure, therefore, it has been adopted by a recent study for assessing the lung cancer risks of general adults exposure to the ambient atmospheric PAHs (Zmirou et al., 2000). For occupational exposure, Pott (1985) established a relationship between BaP exposure and lung cancer risk based on a data bank provided by an epidemiological study conducted by Redmond et al (1976). He suggested a unit risk of  $7.0\times10^{-2}$ 23  $(\mu g/m^3)^{-1}$  for a 25 years occupational PAHs exposure with an averaged BaP concentration of 1  $\mu$ g/m<sup>3</sup>. By using the same data bank, the US EPA, however, suggested a different unit risk of

 6.4×10<sup>-4</sup> ( $\mu$ g/m<sup>3</sup>)<sup>-1</sup> for PAHs exposure based on its total PAH content (expressed as the benzene soluble fractions) (US EPA, 1984). Since a recent study has indicated BaP could be a better indicator than total PAH content on characterizing the carcinogenic potency of PAHs (Petry et al., 1996), the unit risk suggested by Pott (1985) was used in this study. In addition, to directly adopt the above unit risk, the period of 25 years occupational exposure was used in estimating workers' lung cancer risks accordingly.

 To estimate the skin cancer risk posed by 25 years occupational dermal PAHs exposure, a 8 unit risk  $37.47$ (mg/kg bodyweight/day)<sup>-1</sup> suitable for low dose exposures (such as environmental and occupational exposures) was adopted in this study, assuming personal averaged BaP exposure level was 1 mg/kg bodyweight/day (Hussain et al., 1998). The use of the above unit risk in the present study is because the above value was obtained by extrapolating the lifetime risks from high doses to low doses by using the Model-Free Extrapolation (MFX) computer model, assuming a linear dose-response curve at low doses (Krewski et al., 1991).

# **3. Results and Discussion**

#### *3.1 Personal inhalatory PAH exposures*

 In this study, we found that the mean individual and total PAH exposure concentrations of gas phase, particle phase and gas+particle phase for all selected threading workers were log-normally 18 distributed ( $p$ <0.001, W-test). The above result suggests that the selected threading workers can be regarded as a similar exposure group (Mulhausen and Damiano, 1998). Table 3 shows the 20 estimated mean and its corresponding  $LCL_{1.95\%}$  and  $UCL_{1.95\%}$  PAH exposure concentrations of gas phase, particle phase and gas+particle phase for threading workers. We found that the mean 22 gas phase total PAH exposure level  $(8.60\times10^4 \text{ ng/m}^3)$  was much higher than that of particle phase 23 (2.30 $\times$ 10<sup>3</sup> ng/m<sup>3</sup>). By examining PAH homologue distributions for both gas-phase and particle-phase PAHs, it can be seen that the fractions of LMW-, MMW- and HMW-PAHs for the former were 97.2%, 2.5%, and 0.3%, and for the latter were 66.8%, 18.3%, and 14.9%, respectively. Gas-phase PAHs had a higher fraction in LMW-PAHs was because of their intrinsic



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- 25 exposure levels of  $7.38 \times 10^4$ ,  $5.41 \times 10^4$ ,  $3.24 \times 10^4$ ,  $2.48 \times 10^4$ , and  $2.33 \times 10^4$  ng/100cm<sup>2</sup>, respectively.
- 26 Clearly, the above high exposure levels were mainly because their surface areas were uncovered

 (or partly uncovered) by clothes, and they are closer to the emission source than other body surface areas. Here, it should be noted that the exposure level of hand was lower than that of upper arm was because the latter was covered by cotton gloves. On the other hand, we found that those six surface areas covered by clothes (i.e., head/back, chest/abdomen, back, upper leg, lower 5 leg, and foot) had lower exposure levels (range=  $1.34 \times 10^4 - 2.29 \times 10^4$  ng/100cm<sup>2</sup>) than the above 6 five surface areas. The same trend can also be seen in dermal total-BaP<sub>eq</sub> exposure levels for each individual body surface (Table 4).

 In this study we also found that the fractions of LMW-, MMW-, and HMW-PAHs in DE*total,whole* (=70.3%, 21.4%, and 8.3%, respectively; data not shown in Table 4) was quite similar to that of inhalatory particle phase PAH exposures (=68.8%, 18.3%, and 14.9%, respectively; Table 3), but was very different from that of inhalatory gas phase (=97.2%, 2.5%, and 0.3%, respectively; Table 3) and gas +particle phase PAHs (96.4%, 2.9%, 0.7%, respectively; Table 3). The above results suggest airborne particle phase PAHs was the main contributor for dermal PAH exposures. Considering those body surface areas uncovered (or partly uncovered) by clothes had higher exposure levels, it implies that the impaction of PAHs-containing oil mists on the uncovered body surface areas might play the important role on workers' dermal exposures.

# *3.3 Health-risk assessments for both inhalatory and dermal PAH exposures*

18 In this study, workers' total  $BaP_{eq}$  levels were used to estimate the mean, and its

19 corresponding  $LCL_{1.95\%}$  and  $UCL_{1.95\%}$  lifetime lung cancer risks for the studied exposure group.

20 As shown in Table 5, the estimated mean lifetime lung cancer risk  $(1.64 \times 10^{-2})$  was contributed

21 more by particle phase PAH exposures  $(8.81 \times 10^{-3})$  than by gas phase PAH exposures  $(7.55 \times 10^{-3})$ .

However, it should be noted that in this study particle phase PAHs was vigorously extracted by

using a solvent solution (i.e., a mixture of n-hexane and dichloromethane), which could be very

24 different from that could be extracted in the respiratory system. Particularly, it is expected that the

health risk associated with particle phase PAH exposures could be overestimated. At this stage,

the elution rate of particle-phase PAHs deposited in lung still remains unknown which warrants

the need for further investigation in the future.

2 Table 5 also shows the estimated mean (and its corresponding  $LCL_{1.95\%}$  and  $UCL_{1.95\%}$ ) lifetime skin cancer risk for the selected threading workers. To date, the desorption rate of PAHs for oil mists deposited on the skin surface still remains unknown. However, US EPA (1992) has suggested a maximum of 20% of PAHs that adsorbed onto soil could be desorbed onto the skin in 24 hours. The above value was adopted in this study. Assuming the body weight of the standard man (=70 kg) is representative to all threading workers and all of them will be exposed to oil mists for 25 years (5 days per week, 50 weeks per year) during an average life span of 70 years. 9 The resultant mean and its corresponding LCL<sub>195%</sub> and UCL<sub>195%</sub> lifetime dermal total BaP<sub>eq</sub> 10 exposure levels for the threading workers were  $2.59 \times 10^{-1}$ ,  $1.77 \times 10^{-1}$ , and  $4.48 \times 10^{-1}$  µg/kg/day, 11 respectively. The unit risk  $(37.47 \times 10^{-3} (\mu g/kg / day)^{-1})$  suggested by Hussain et al. (1998) was adopted in this study to estimate the lifetime skin cancer risk. The estimated mean and its 13 corresponding LCL<sub>1,95%</sub> and UCL<sub>1,95%</sub> lifetime skin cancer risks were  $9.72 \times 10^{-3}$ , 6.64 $\times 10^{-3}$ , and 14 1.68 $\times$ 10<sup>-2</sup>, respectively. Although the resultant mean lifetime skin cancer risks (9.72 $\times$ 10<sup>-3</sup>) was 15 slightly lower than that of the mean lifetime lung cancer risk  $(1.64 \times 10^{-2})$ , both are higher than the 16 significant risk level  $(10^{-3})$  defined by the US Supreme Court in 1980. Therefore, it is recommended that measurements should be taken in order to reduce both inhalatory and dermal exposures.

 Considering PAHs contained oil mists contributed to not only inhalatory but also dermal exposures, the installation of local exhaust ventilation systems at the emission source (i.e., the threading machine) should be considered as the first priority measurement because both lung and skin cancer risks can be reduced simultaneously. On the other hand, if the personal protection equipment will be adopted for exposure abatement purpose, both respiratory protection equipment and protective clothing should be used simultaneously from the health risk management aspect.

# **4. Conclusions**

We found that the inhalatory gas-phase total PAHs exposure level was much higher than that

 of particle-phase. By combining both gas- and particle-phase total PAHs, workers' mean 2 inhalatory exposure level was lower, but its corresponding  $UCL_{1,95\%}$  was slightly higher than the PEL-TWA for PAHs. On the other hand, in our previous study we found that workers' mean oil mist exposure concentration was much lower than the current PEL-TWA for oil mist. The present study clearly indicates the inadequacy in assessing workers' inhalatory exposures if only oil mist concentrations were measured. For workers' dermal exposures, the top five exposed surface areas were lower arm, hand, upper arm, neck, and head/front mainly because their surface areas were uncovered (or partly uncovered) by clothes. The estimated lifetime lung cancer risk caused by inhalatory PAH exposures was higher than that of skin cancer risks caused by dermal exposures. However, both cancer risks are higher than the significant risk level defined by the US Supreme Court in 1980. Therefore, it is recommended that measurements should be taken in order to reduce both inhalatory and dermal exposures. The installation of local exhaust ventilation systems at the emission source should be considered as the first priority measurement because both lung and skin cancer risks can be reduced simultaneously. If the personal protection equipment would be adopted in the future, both respiratory protection equipment and protective clothing should be used simultaneously from the health risk management aspect. **Acknowledgment** The authors wish to thank National Science Council in Taiwan for funding this research project.

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# **Table Captions**

