

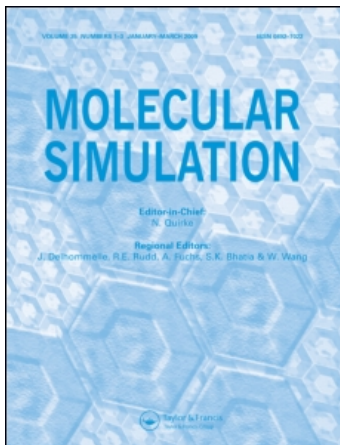
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### Structure-based and ligand-based drug design for microsomal prostaglandin E synthase-1 inhibitors

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## Structure-based and ligand-based drug design for microsomal prostaglandin E synthase-1 inhibitors

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Microsomal prostaglandin E synthase-1 (mPGES-1) has been regarded as an attractive drug for inflammation-related diseases. In search of new mPGES-1 inhibitors, we performed virtual screening using our traditional Chinese medicine and natural products database (<http://tcm.cmu.edu.tw/>) and constructed comparative molecular field analysis (CoMFA) and comparative molecular similarity indices analysis (CoMSIA) using a training set of 30 experimentally tested mPGES-1 inhibitors. The CoMFA and CoMSIA models derived were statistically significant with cross-validated coefficient values of 0.808 for CoMFA and 0.829 for CoMSIA and non-cross-validated coefficient values of 0.829 for CoMFA and 0.980 for CoMSIA. Docking and *de novo* evolution design gave three top derivatives, 2-O-caffeoyl tartaric acid-Evo\_2, glucogallin-Evo\_1 and 3-O-feruloylquinic acid-Evo\_7 that have higher binding affinities than the control, glutathione. These three derivatives have interactions with Arg70, Arg73, Arg110, Arg126 and Arg38, which all are mPGES-1 key active site residues. In addition, these derivatives fit well into the CoMFA and CoMSIA models, with hydrophobic, hydrophilic and electropositive substructures mapped onto corresponding contour plots. Hence, we suggest that these three *de novo* compounds could be a starting basis for new mPGES-1 inhibitors.

**Keywords:** microsomal prostaglandin E synthase-1; QSAR; docking traditional Chinese medicine; database

### 1. Introduction

In the prostaglandin (PG) biosynthetic pathway, arachidonic acid is converted to PG H<sub>2</sub> (PGH<sub>2</sub>) by cyclooxygenase (COX). PGE<sub>2</sub> synthase catalyses the conversion of PGH<sub>2</sub> to PG E<sub>2</sub> (PGE<sub>2</sub>), the most abundant PG in the body. PGE<sub>2</sub> is an important product of the COX pathway and has been considered as a mediator of inflammation, pain, fever and cancer [1,2]; it is also known to regulate physiological functions in the gastrointestinal tract, in the kidney and in the immune and nervous systems.

Three major isoforms of PGE<sub>2</sub> synthases have been identified: cytosolic PG E synthase (cPGES), microsomal PGES 1 (mPGES-1) and mPGES-2 [3,4]. Both cPGES and mPGES-2 are constitutively expressed in various cells and tissues, while mPGES-1 is localised to microsomal compartment of the cell. Recently, mPGES-1 has attracted

much attention as a potential drug target for inflammation and pain [5], tumorigenesis, arthritis and atherosclerosis [6], stroke [7,8], cancer [9,10] and tissue repair [11].

Inflammations are related to many diseases, including cancer, arthritis and atherosclerosis. Specifically, mPGES-1 is induced by pro-inflammatory agents and is functionally coupled to COX-2 in various models of inflammation. Previous studies using mPGES-1 knockout mice have shown that mPGES-1 contributes to the inflammatory production of PGE<sub>2</sub>. Hence, mPGES-1 is a potential drug target for inflammation-related diseases.

Previously, we have investigated mPGES-1 pharmacophore features by Catalyst HypoGen [12], and as a continuation of the project, we applied 3D quantitative structure–activity relationship (3D-QSAR) analysis and molecular docking to search for novel inhibitors for mPGES-1. Both 3D-QSAR and docking techniques are

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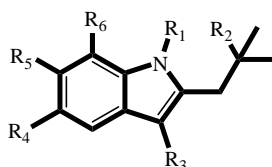
† All these authors contributed equally to this work.

disciplines of computer-aided drug design (CADD) that have been used in pharmaceutical industry and in our lab for drug development process [13–26].

Aside from employing CADD technologies, we introduced a traditional Chinese medicine (TCM) and natural product database (<http://tcm.cmu.edu.tw/>) in the virtual screening process. Herbal medicines and

natural products have been used in many cultures for hundreds of years, and some have been recently studied to identify the biological activity of the principal components [27–36]. Hence, our goal is to identify natural compounds that possess potential inhibitory activity towards mPGES-1 and that could be further modified for better pharmacological activity.

Table 1. Molecular structure of the training set and the test set



Compound	R1	R2	R3	R4	R5	R6
Training set						
1	CH2(4-Cl-Ph)	COO-	S-tBu	iPr		H
2	H	COO-	S-tBu	iPr		H
3	Me	COO-	S-tBu	iPr		H
4	CH2(CHdCH2)	COO-	S-tBu	iPr		H
5	(CH2)3Ph	COO-	S-tBu	iPr		H
6	CH2(4-Cl-Ph)	CO2Me	S-tBu	iPr		H
7	CH2(4-Cl-Ph)	CONH2	S-tBu	iPr		H
8	CH2(4-Cl-Ph)	COO-	Ph	iPr		H
9	CH2(4-Cl-Ph)	COO-	OPh	iPr		H
10	CH2(4-Cl-Ph)	COO-	CH2(4-tBu-Ph)	iPr		H
11	CH2(4-Cl-Ph)	COO-	CO(2-Me-Ph)	iPr		H
12	CH2(4-Cl-Ph)	COO-	COCH2S-tBu	iPr		H
13	CH2(4-Cl-Ph)	COO-	COCH2-tBu	iPr		H
14	CH2(4-Cl-Ph)	COO-	Me	iPr		H
15	CH2(4-Cl-Ph)	COO-	Me	H		iPr
16	CH2(4-Cl-Ph)	COO-	Me	H		H
17	CH2(4-Cl-Ph)	COO-	Me	F		H
18	CH2(4-Cl-Ph)	COO-	Me	tBu		H
19	CH2(4-Cl-Ph)	COO-	Me	Ph		H
20	CH2(4-Cl-Ph)	COO-	Me	(3-Ph)-Ph		H
21	CH2(4-Cl-Ph)	COO-	Me	(4-Ph)-Ph		H
22	CH2(4-Cl-Ph)	COO-	Me	(3-Cl,4-Ph)-Ph		H
23	CH2(4-Cl-Ph)	COO-	Me	(3-F,4-Ph)-Ph		H
24	CH2(4-Cl-Ph)	COO-	Me	[3-F,4-(2-MeO-Ph)]-Ph		H
25	CH2(4-Cl-Ph)	COO-	Me	[3-F,4-(2-Cl-Ph)]-Ph		H
26	CH2(4-Cl-Ph)	COO-	Me	[3-F,4-(2-F-Ph)]-Ph		H
27	CH2(4-Cl-Ph)	COO-	Me	[3-F,4-(2-MeCO-Ph)]-Ph		H
28	CH2(4-Cl-Ph)	COO-	Me	[3-F,4-(2-Me-Ph)]-Ph		H
29	CH2(4-Cl-Ph)	COO-	Me	[3-F,4-(3-Me-Ph)]-Ph		H
30	CH2(4-Cl-Ph)	COO-	Me	[3-F,4-(4-Me-Ph)]-Ph		H
Test set						
31	CH2(4-Cl-Ph)	CO2H	COCH2S-tert-Bu	2-Methyl		H
32	CH2(4-Cl-Ph)	CO2H	COCH2-tert-Bu	2-Methyl		H
33	CH2(4-Cl-Ph)	CO2H	Me	2-Methyl		H
34	CH2(4-Cl-Ph)	CO2H	Me	H		Iso-propyl
35	CH2(4-Cl-Ph)	CO2H	Me	tert-butyl		H
36	CH2(4-Cl-Ph)	CO2H	Me	Ph		H
37	CH2(4-Cl-Ph)	CO2H	Me	Ph		H
38	CH2(4-Cl-Ph)	CO2H	Me	Ph	F	H
39	CH2(4-Cl-Ph)	CO2H	Me	2-MeO-Ph	F	H
40	CH2(4-Cl-Ph)	CO2H	Me	2-Cl-Ph	F	H
41	CH2(4-Cl-Ph)	CO2H	Me	2-MeCO-Ph	F	H
42	CH2(4-Cl-Ph)	CO2H	Me	3-Me-Ph	F	H

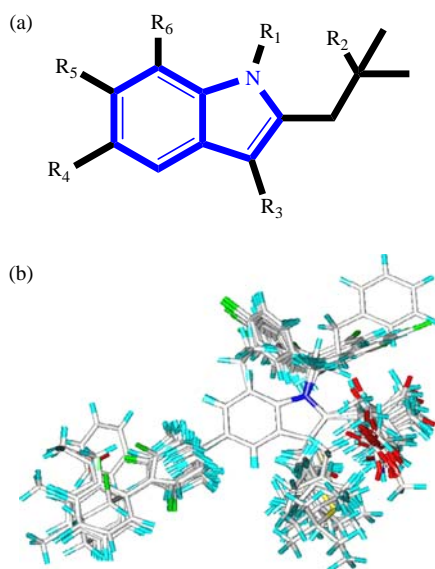


Figure 1. (a) Alignment scaffold for the training set and test set. The core atoms are labelled in bold face blue and (b) the alignment of training set molecules (colour online).

## 2. Materials and method

### 2.1 3D quantitative structure–activity relationship

We used SYBYL7.3 (Tripos Inc., St. Louis, MO, USA) to build 3D-QSAR models. Forty-two mPGES-1 inhibitors

used in our study were taken from previous literatures [37,38]. The structures of 42 compounds are shown in Table 1. The 2D and 3D structures of these compounds were drawn using ChemBioOffice 2008 (CambridgeSoft Inc., Cambridge, MA, USA) (compound structures shown in Table 1). Structure optimisation was carried out using the MM2 force field. These 42 inhibitors were further divided into the training set and the test set.

The steric and electrostatic energies investigated in comparative molecular field analysis (CoMFA) were calculated by using a carbon probe and a positive charge. Hydrophobic and hydrogen bond donor and acceptor fields were computed in comparative molecular similarity indices analysis (CoMSIA) in addition to steric and electrostatic fields.

### 2.2 Partial least squares

Partial least squares (PLS) analyses were utilised to analyse the 3D-QSAR models. All CoMFA and CoMSIA fields were regarded as independent variables. By using the leave-one-out method, the value of cross-validation coefficient,  $q^2$ , was further employed for non-cross validation to acquire the highest non-cross validation coefficient,  $r^2$ , the lowest standard error of estimate, and  $F$ -value. The final analysis of the non-cross validation was used to predict the test set activity and to derive the contour maps of CoMFA and CoMSIA.

Table 2. PLS statistics of CoMFA and CoMSIA.

	CoMFA	CoMSIA	Cross validation		Non-cross validation		
			ONC	$q^2_{cv}$	$r^2$	SEE	$F$
ONC	4	A	5	0.052	0.414	1.008	3.391
$q^2_{cv}$	0.808	E	5	0.466	0.909	0.398	47.715
SEE	0.199	H	6	0.790	0.996	0.088	890.823
$r^2$	0.976	S	6	0.819	0.976	0.210	154.122
$F$	255.493	A + D	4	0.077	0.359	1.043	3.493
		A + E	4	0.574	0.900	0.409	56.098
		A + H	5	0.822	0.967	0.241	138.623
		A + S	4	0.818	0.936	0.326	91.982
		E + S	5	0.726	0.969	0.231	151.323
		H + S	6	0.818	0.996	0.082	1028.592
		A + D + E	6	0.475	0.947	0.310	68.447
		A + D + H	6	0.799	0.973	0.219	140.365
		A + D + S	5	0.818	0.938	0.327	73.193
		A + E + S	5	0.743	0.968	0.234	147.395
		A + E + H	6	0.721	0.982	0.181	207.163
		A + H + S	5	0.829	0.980	0.187	232.362
		D + E + S	4	0.668	0.937	0.325	92.306
		E + H + S	4	0.784	0.977	0.196	265.398
		A + D + E + H	5	0.680	0.968	0.234	146.716
		A + D + E + S	5	0.702	0.959	0.267	112.268
		A + D + H + S	4	0.808	0.948	0.293	114.981
		A + E + H + S	6	0.793	0.988	0.145	327.459
		A + D + H + E + S	6	0.761	0.986	0.161	264.192

Notes: ONC, optimal number of component; SEE, standard error of estimate;  $F$ ,  $F$ -test value; PLS, partial least squares; S, steric; E, electrostatic; H, hydrophobic; D, hydrogen bond donor; A, hydrogen bond acceptor

### 2.3 Docking

Our TCM and natural compound database (<http://tcm.cmu.edu.tw/index.php>) were used in the docking process. All compounds were drawn using ChemBioOffice 2008 (CambridgeSoft Inc., Cambridge, MA, USA) and energetically minimised in the MM2 forcefield.

All molecular simulations were performed by Discovery Studio 2.5 (Accelrys Inc., San Diego, CA, USA) and CHARMM (Chemistry at Harvard macromolecular mechanics). The protein structure of mPGES-1 was obtained from Protein Data Bank (PDB ID: 3DWW) [39] and employed in the docking study with our laboratory database. The docking procedure first defined the ligand binding site in a receptor

before docking ligands into the specified site. LigandFit docking program and scoring functions (DockScore, piecewise linear potential (PLP), potential of mean force (PMF) and LigScore) were used to evaluate the receptor–ligand binding affinity of all compounds and their possible poses. PF-9184 [40], 4-benzo[*b*]thiophen-2-yl-3-bromo-5-hydroxy-5*H*-furan-2-one (BTH) [41] and glutathione were taken as controls for filtering molecules.

### 2.4 Ligand de novo evolution

In *de novo* evolution, derivatives were generated based on existing scaffolds. The Ludi algorithm was used to

Table 3. CoMFA and CoMSIA prediction results of mPGES-1.

Compound	CoMFA		CoMSIA		
	pIC <sub>50</sub> exp	pIC <sub>50</sub> pred	Residual	pIC <sub>50</sub> pred	Residual
1	5.796	5.793	0.003	5.835	-0.039
2	5.000	4.944	0.056	4.781	0.219
3	5.000	4.901	0.099	4.981	0.019
4	5.174	5.146	0.028	5.297	-0.123
5	5.495	5.688	-0.193	5.671	-0.176
6	5.143	5.334	-0.191	5.068	0.075
7	5.000	5.195	-0.195	4.897	0.103
8	5.194	5.350	-0.156	5.397	-0.203
9	6.187	5.676	0.511	6.156	0.031
10	6.538	6.651	-0.113	6.561	-0.023
11	6.046	6.047	-0.001	6.166	-0.120
12	6.585	6.554	0.031	6.787	-0.202
13	6.602	6.620	-0.018	6.634	-0.032
14	5.959	6.056	-0.097	5.957	0.002
15	5.367	5.175	0.192	5.184	0.183
16	5.495	5.577	-0.082	5.645	-0.150
17	5.585	5.591	-0.006	5.601	-0.016
18	6.481	6.213	0.268	6.109	0.372
19	6.222	6.285	-0.063	6.302	-0.080
20	6.796	6.780	0.016	6.796	0.000
21	7.796	8.101	-0.305	8.009	-0.213
22	7.658	7.897	-0.239	7.662	-0.004
23	8.155	8.183	-0.028	8.191	-0.036
24	8.301	8.101	0.201	7.980	0.321
25	8.398	8.125	0.273	8.050	0.348
26	8.097	8.072	0.025	8.093	0.004
27	8.222	8.194	0.028	8.324	-0.102
28	8.523	8.198	0.325	8.273	0.250
29	7.481	7.654	-0.173	7.638	-0.157
30	7.509	7.704	-0.195	7.760	-0.251
31	6.585	6.152	0.433	6.655	-0.070
32	6.602	6.839	-0.237	6.788	-0.186
33	5.959	6.600	-0.641	6.455	-0.496
34	5.367	5.580	-0.213	5.286	0.081
35	6.481	6.996	-0.515	6.899	-0.418
36	6.222	6.737	-0.515	6.486	-0.264
37	6.796	6.938	-0.142	6.914	-0.118
38	8.155	8.209	-0.054	8.254	-0.099
39	8.301	8.245	0.056	8.250	0.051
40	8.398	8.639	-0.241	8.411	-0.013
41	8.222	8.217	0.005	8.174	0.048
42	7.481	6.999	0.482	6.803	0.678

Notes: pIC<sub>50</sub> exp is experimental pIC<sub>50</sub>; pIC<sub>50</sub> pred is predicted pIC<sub>50</sub>.



evaluate fragments that best complement the receptor. These fragments are fused onto the molecule scaffold, generating a collection of molecules with high Ludi score. Top candidates from previous docking study were taken for *de novo* evolution, and the resultant derivatives were first screened by the Lipinski's rule of five before being docked back to the mPGES-1 active site again for evaluating binding poses and binding affinity.

### 3. Results and discussion

#### 3.1 3D-QSAR analysis

In this study, we used a training set of 30 compounds to develop our CoMFA and CoMSIA models. All 30 compounds were aligned to core atoms shown in blue in

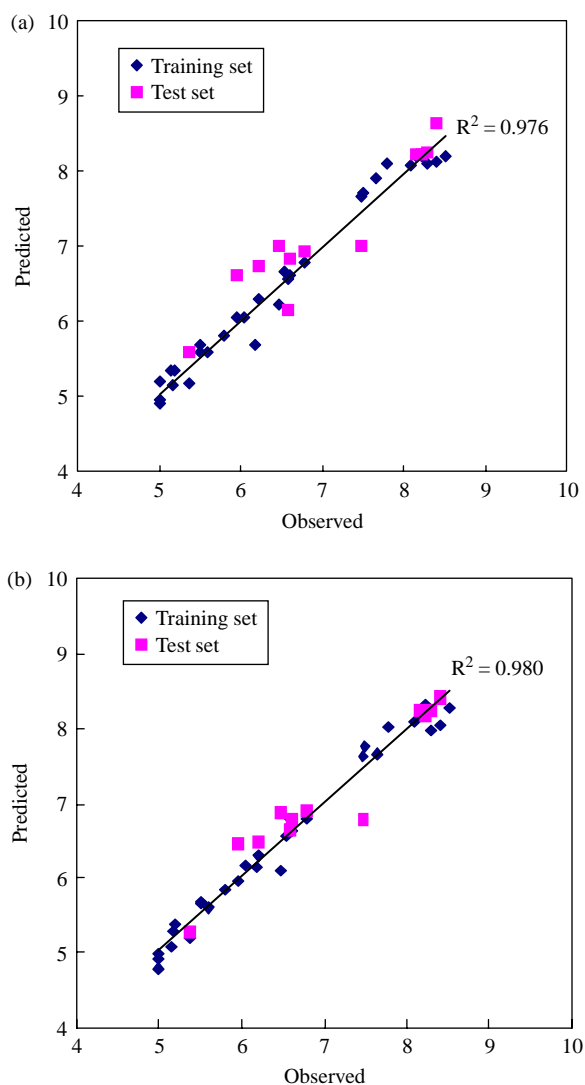


Figure 2. Plot of observed activities versus predicted activities for the training (◆) and test set (■) compounds based on prediction of (a) CoMFA and (b) CoMSIA.

Figure 1(a). The alignment of the training set molecules is shown in Figure 1(b).

PLS analyses of the training set gave a cross-validated coefficient ( $q^2$ ) value of 0.808, a non-cross-validated coefficient ( $r^2$ ) of 0.976 and a standard error of estimate (SEE) value of 0.199 for CoMFA (Table 2). As for CoMSIA, the best model (A + H + S) showed a  $q^2$  value of 0.829, an  $r^2$  value of 0.980 and an SEE value of 0.187 (Table 2). The high  $r^2$  and  $q^2$  values and the low SEE value indicate that the 3D-QSAR models are reasonable and should have a good predictive ability.

The predictive ability of the 3D-QSAR models was further verified with an external test set of 12 compounds. The predicted activities of the training and test set molecules were compared with the experimental obser-

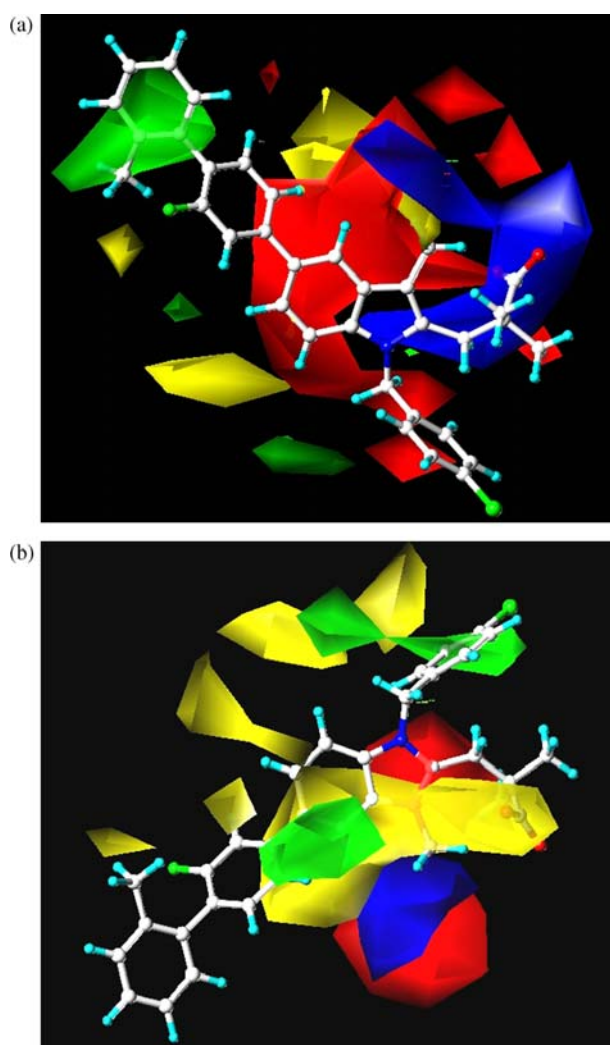


Figure 3. (a) CoMFA contour map. Steric field: favour (green) and disfavour (yellow). Electropositive field: favour (blue) and disfavour (red) and (b) CoMSIA contour map. Hydrophobic field, favour (green) and disfavour (yellow); hydrogen bond acceptor, favour (white) and disfavour (blue); hydrogen bond donor, favour (red) and disfavour (purple) (colour online).

vations (Table 3). As illustrated in Figure 2, the predicted activities for both training and test set compounds showed high correlation with the experimental activities and gave a correlation value ( $R^2$ ) of 0.976 for CoMFA and 0.980 for CoMSIA. These statistical results suggest that both CoMFA and CoMSIA have good predictions for the training and the test set compounds.

### 3.2 Contour map analyses of CoMFA

The contour maps of CoMFA and CoMSIA models are shown in Figure 3. The most active training set compound (**28**) is superimposed onto the CoMFA and CoMSIA contours for references. In the CoMFA model (Figure 3(a)), the steric contour maps are represented in green and yellow for favoured and disfavoured contribution, respectively. The large green contour region surrounding the terminal phenyl ring (R4 group) suggests that the steric bulk group could enhance biological activity in this region. This is in good agreement with the experimental results for compounds **19–30**, which all have a terminal phenyl group in this region and all have  $IC_{50}$  less than 40 nM. The electrostatic contour maps are shown in red for electronegative contribution and in blue for electropositive contribution. The red contours are found surrounding the terminal phenyl group (R1 group), suggesting that an addition of electronegative substituents could increase biological activity. On the other hand, the blue contours are concentrated near the carboxylate group (R2 group), favouring modification to electropositive group.

### 3.3 Contour map analyses of CoMSIA

In CoMSIA model, hydrophobic and hydrogen bonding interaction were included in addition to steric and electrostatic field descriptors. In Figure 3(b), the hydrophobic contour maps are shown in green or in yellow for favoured or disfavoured contribution. The hydrogen bond acceptor contour maps are indicated in white for favoured contribution and in blue for the opposite. The contour maps are shown in red or in purple for region favouring or disfavouring hydrogen bond donor. The green contour near the terminal phenyl group (R1 group) indicates that a hydrophobic group in this region is favourable. The yellow contour region near the carboxylate group (R2 group) indicates that a hydrophilic group is favoured. Hence, to enhance the compound activity, we could introduce hydrophobic groups to the terminal phenyl group found at the R1 position and hydrophilic groups to the carboxylate group at the R2 position.

### 3.4 Docking and de novo evolution analysis

In the docking study, the binding affinity between ligands and the receptor was calculated using scoring functions. The DockScores for controls, PF-9184, BTH and glutathione, were 53.466, 54.757 and 66.787, respectively. Preliminary docking experiments resulted in 81 compounds that have DockScores higher than the controls (Table 4; only the top 20 compounds are shown). The top three candidates are 2-O-caffeoyl tartaric acid, chicoric

Table 4. Docking results.

Compound	D.S	PLP1	PLP2	PMF	LigS1	LigS2
2-O-Caffeoyl tartaric acid	215.079	68.06	90.04	224.93	6.64	5.88
Chicoric acid	206.092	117.44	123.46	256.74	7.05	6.05
Mumefural	201.985	72.59	87.65	198.04	6.23	4.96
2-O-Feruloyl tartaric acid	198.739	70.84	83.36	224.55	6.19	5.13
Salvianolic acid B	156.923	110.76	123.01	133.86	7.59	6.83
Rosmarinic acid	148.434	81.92	101.21	223.61	6.79	5.88
Quinic acid	143.961	49.97	60.81	164.90	5.28	4.62
Genipinic acid	142.772	53.24	57.96	195.62	5.65	5.05
Digallic acid	142.547	64.37	81.51	207.53	6.35	5.57
5-O-Feruloylquinic acid	142.462	90.73	101.31	225.85	6.77	5.73
3-O-Feruloylquinic acid	140.488	91.52	106.21	218.68	6.98	5.86
Lithospermic acid	140.305	73.86	111.21	243.55	5.94	3.45
Isochlorogenic acid	140.272	71.80	80.80	228.69	6.45	5.47
Gallic acid	140.219	44.47	53.19	119.23	4.45	4.16
2, 3-Dihydroxycinnamic acid	139.170	42.33	52.25	121.93	4.92	4.29
1-Caffeoylquinic acid	139.057	86.48	92.77	227.00	6.19	5.16
Protocatechuic acid	137.016	38.80	45.62	119.88	4.29	4.18
Shikimic acid	136.811	43.34	50.65	137.44	4.51	4.16
4-O-Caffeoylquinic acid	136.713	81.41	100.12	208.58	6.68	5.50
5-O-Caffeoylshikimic acid	136.493	72.33	89.63	184.15	5.66	4.65
PF-9184	53.466	93.82	100.17	175.53	5.27	6.21
BTH	54.757	61.90	59.57	103.39	4.39	4.76
Glutathione	66.787	102.17	98.17	193.40	7.02	6.28

Notes: D.S, Dock score; LigS1, LigScore1; LigS2, LigScore2; only the top 20 candidates and controls are shown.

Table 5. Docking results for *de novo* products.

Compound	D.S	PLP1	PLP2	PMF	LigS1	LigS2
2-O-Caffeoyl tartaric acid-Evo_2	222.198	87.60	98.80	196.39	6.24	5.43
Glucogallin-Evo_1	169.762	86.96	107.38	243.42	7.03	6.14
3-O-Feruloylquinic acid-Evo_7	167.056	95.10	111.99	234.69	7.13	6.09
1-Caffeoylquinic acid-Evo_3	165.916	94.30	101.14	240.61	6.62	5.12
3-O-Feruloylquinic acid-Evo_5	159.929	96.61	116.46	250.69	7.27	5.99
Capillartemisin B-Evo_4	152.081	90.53	101.68	216.51	6.28	4.17
Capillartemisin B-Evo_1	151.803	94.49	106.10	222.5	6.30	4.02
Capillartemisin B-Evo_6	150.856	84.49	95.74	213.4	6.25	4.14
Gentisic acid-Evo_1	146.487	57.94	60.68	151.53	4.90	4.83
3-O-Feruloylquinic acid-Evo_2	145.552	87.65	97.85	248.49	6.17	5.08
O-Coumaric acid-Evo_2	145.117	63.83	71.65	155.32	5.64	4.98
Capillartemisin B-Evo_5	144.76	87.00	94.31	203.15	5.46	3.84
Capillartemisin B-Evo_2	144.719	87.51	96.51	200.48	5.49	3.83
O-Coumaric acid-Evo_1	144.069	64.43	73.19	148.43	5.64	5.04
Picrocrocinic acid-Evo_1	141.492	93.51	110.13	242.07	6.18	3.96
3-O-Feruloylquinic acid-Evo_3	141.444	86.44	99.14	237.14	7.15	5.86
O-Coumaric acid-Evo_5	141.085	58.98	65.56	142.49	5.46	4.76
Caffeic acid-Evo_1	140.641	37.91	52.82	133.02	4.88	4.63
3,5-Dihydroxycinnamic acid-Evo_1	139.832	37.91	52.82	131.04	4.88	4.63
Ferulic acid-Evo_5	139.070	56.55	67.36	151.75	5.91	5.16

Note: The top 20 candidates are shown here.

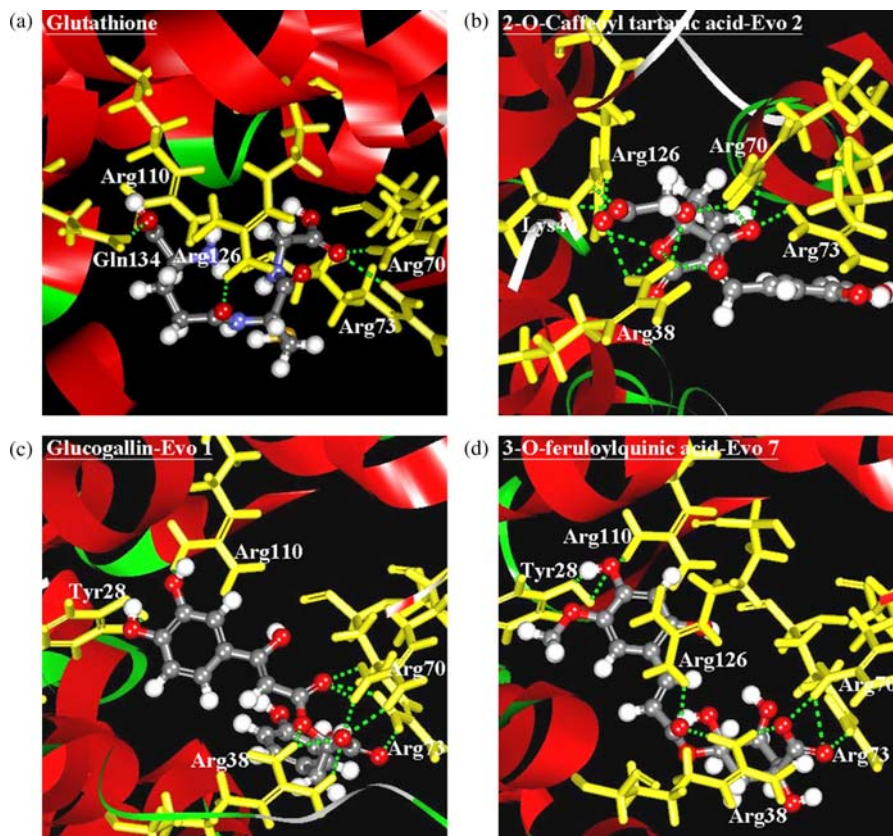


Figure 4. Binding conformation of (a) Glutathione, (b) 2-O-Caffeoyl tartaric acid-Evo\_2, (c) Glucogallin-Evo\_1 and (d) 3-O-Feruloylquinic acid-Evo\_7 at the mPGES-1 binding site.



Table 6. Number of hydrogen bonds formed between ligands and mPGES-1 active site residues.

Compound	Arg70	Arg73	Arg110	Arg126	Gln134	Tyr28	Arg38	Lys42
Glutathione	1	1	1	1	1	–	–	–
2-O-Caffeoyl tartaric acid-Evo_2	3	1	–	3	–	–	5	2
Glucogallin-Evo_1	4	2	1	–	–	2	3	–
3-O-Feruloylquinic acid-Evo_7	2	1	1	1	–	2	2	–

acid and mumeferal, and their DockScores are 215.079, 206.092 and 201.985, respectively.

A total of 408 derivatives were generated based on the top compounds obtained from the previous docking step. Only 50 derivatives that passed Lipinski's rule of five were docked back to the mPGES-1 binding site. The top 20 *de novo* compounds, ranked in order of their DockScores, are shown in Table 5. The top three *de novo* compounds are 2-O-caffeoyl tartaric acid-Evo\_2, glucogallin-Evo\_1 and 3-O-feruloylquinic acid-Evo\_7, and their DockScores are 222.198, 169.762 and 167.056, respectively.

The binding conformation of glutathione (control) and the top three derivatives are illustrated in Figure 4, which shows ligands forming hydrogen bond interactions with residues Arg70, Arg73, Arg110, Arg126 and Arg38. All these residues have been identified as key active site residues in previously published studies [42–45]. A summary of the number of hydrogen bonds formed between the ligands and mPGES-1 residues is shown in Table 6. All the top three derivatives have greater number of hydrogen bonds than the control, thus providing a possible explanation for their elevated DockScores.

### 3.5 CoMFA and CoMSIA mapping

The top three derivatives mapped onto the CoMFA and CoMSIA models are shown in Figures 5 and 6. The common features of the three derivatives are shown in Figure 7, with red circles indicating characteristics noticed from CoMFA and blue circles for features obtained from CoMSIA. Red circle 1 indicates the site having electropositive features, and red circle 2 indicates the region favouring steric bulk group. Blue circle 1 indicates the region having hydrophobic features, and blue circle 2 represents the site favouring hydrophilic features. As shown in the docking results, all substructures circled by blue circle 2 could form strong hydrogen bonding interactions with the receptor. Therefore, the presence of hydrophilic groups could be a big influence for mPGES-1 inhibitor activity. In addition, the hydrophobic substructure of these three derivatives (circled by blue circle 1) could form hydrophobic interactions with other surrounding residues which further strengthen binding interactions. Based on these findings, these functional groups are suggested to be key factors in designing mPGES-1 inhibitors.

## 4. Conclusions

Several novel anti-cancer or anti-inflammatory compounds were investigated from TCM [46–48]. So, we introduced a TCM database (<http://tcm.cmu.edu.tw/>) into our research. Derivative, 2-O-Caffeoyl tartaric acid-Evo\_2, has the highest DockScore of the 50 *de novo*

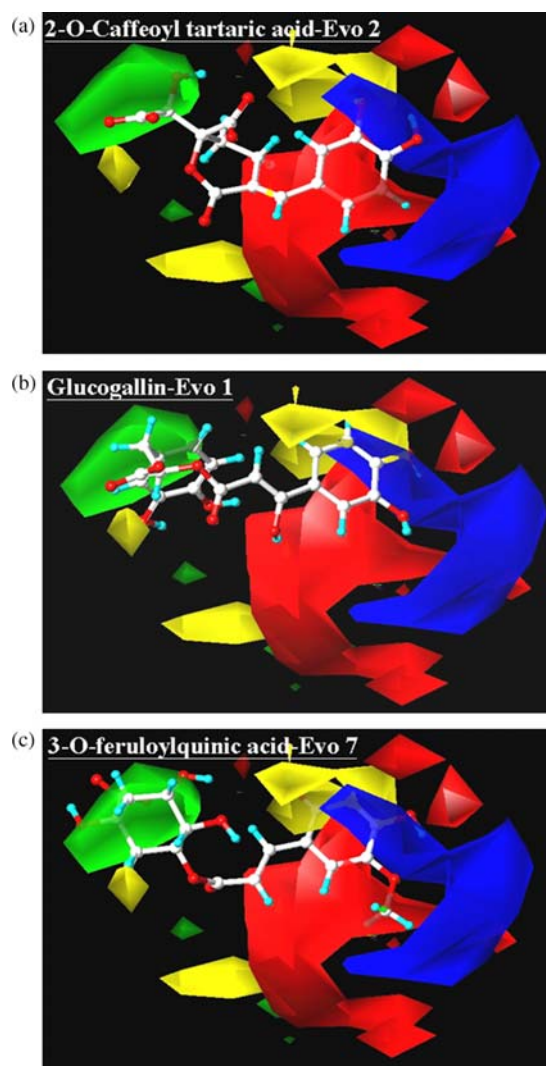


Figure 5. Mapping of *de novo* compounds to CoMFA model. The selected compounds are (a) 2-O-Caffeoyl tartaric acid-Evo\_2, (b) Glucogallin-Evo\_1 and (c) 3-O-Feruloylquinic acid-Evo\_7. Steric field, favour (green) and disfavour (yellow); electropositive field, favour (blue) and disfavour (red) (colour online).

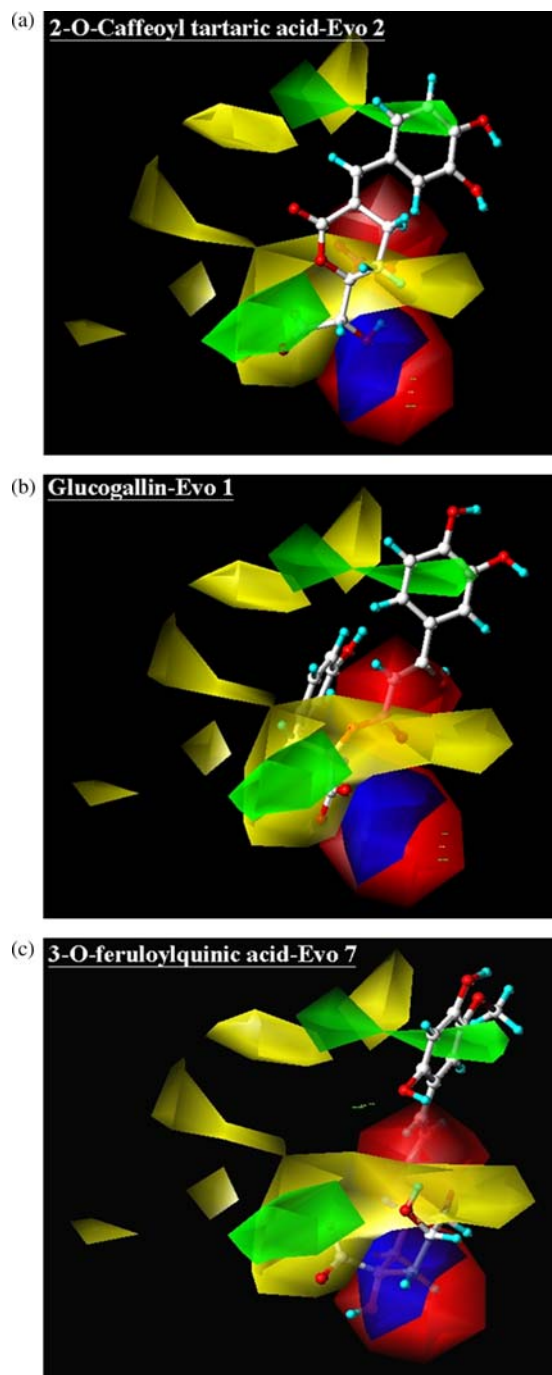


Figure 6. Mapping of *de novo* compounds to CoMSIA model. The selected compounds are (a) 2-O-Caffeoyl tartaric acid-Evo\_2, (b) Glucogallin-Evo\_1 and (c) 3-O-Feruloylquinic acid-Evo\_7. Hydrophobic field, favour (green) and disfavour (yellow); hydrogen bond acceptor, favour (white) and disfavour (blue); hydrogen bond donor, favour (red) and disfavour (purple) (colour online).

products, and its original compound (2-O-Caffeoyl tartaric acid) also ranks first in the preliminary docking. The top three derivatives, 2-O-caffeoyl tartaric acid-Evo\_2, glucogallin-Evo\_1 and 3-O-feruloylquinic acid-Evo\_7,

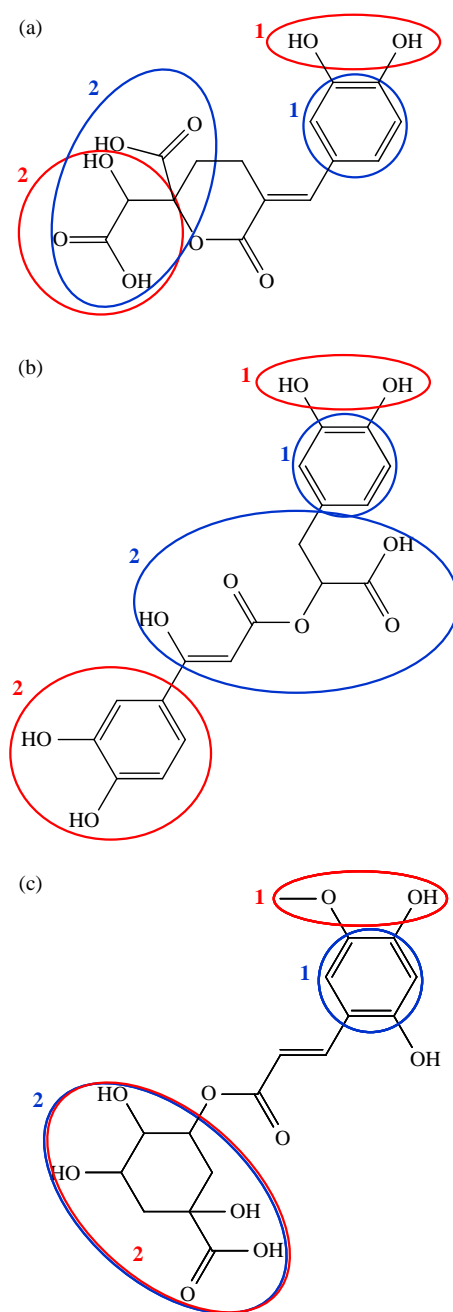


Figure 7. Common features observed among the top derivatives: (a) 2-O-Caffeoyl tartaric acid-Evo\_2, (b) Glucogallin-Evo\_1 and (c) 3-O-Feruloylquinic acid-Evo\_7. Red circle 1 indicates electropositive characteristics; red circle 2 indicates steric favour region. Blue circle 1 indicates hydrophobic features; blue circle 2 indicates hydrophilic features. The red circles are features observed from CoMFA ligand mapping, while the blue circles are obtained from CoMSIA ligand mapping (colour online).

interact with five key mPGES-1 residues, Arg70, Arg73, Arg110, Arg126 and Arg38, respectively. These derivatives mapped well onto the CoMFA and CoMSIA models, and four common chemical features could be observed

from these three derivatives. Overall, we hope that the above findings will provide a constructive idea for designing mPGES-1 inhibitors.

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