## Crassocolides A-F, Cembranoids with a *trans*-Fused Lactone from the Soft Coral *Sarcophyton* crassocaule

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Six new polyoxygenated cembrane-based diterpenoids possessing a *trans*-fused  $\alpha$ -methylene- $\gamma$ -lactone, crassocolides A–F (1–6), have been isolated along with the known compound 7 from the ethyl acetate extract of a Taiwanese soft coral, *Sarcophyton crassocaule*. The structures of 1–6 have been established by detailed spectroscopic analysis, including 2D NMR ( $^{1}$ H– $^{1}$ H COSY, HMQC, HMBC, and NOESY) spectroscopy, while their absolute configurations were determined using a modified Mosher's method for 1. The structure of 5 was further proven by X-ray diffraction analysis. The full assignment of NMR data of 7 is reported herein for the first time. The cytotoxicity of crassocolides 1–4, 6, and 7 against a limited panel of cancer cells was also determined.

Marine terpenoids are of considerable interest due to their unique structures and wide range of biological activities. 1 The macrocyclic cembrane-type compounds have been found to be the main terpenoidal constituents in marine coelenterates.<sup>1</sup> Cembranoids possessing a  $\gamma$ -lactone<sup>2-17</sup> have been mostly isolated from octocorals (Alcyonaceae) belonging to the genera Sinularia, 2-5 Eunicea, 6-12 and Lobophytum. 13-17 Some of these metabolites have been shown to exhibit cytotoxic activity against the growth of various cancer cell lines. <sup>2,6–8,14,15</sup> In our continuing search for bioactive metabolites from soft corals of Taiwanese waters, 18-22 we have chemically investigated Sarcophyton crassocaule Moser and have succeeded in the isolation of six new polyoxygenated trans-fused cembranolides, crassocolides A-F (1-6), along with the known metabolite  $7^{23}$  from the EtOAc extract of the organism. The structures of 1-6have been established by extensive spectroscopic analysis, including 2D NMR (1H-1H COSY, HMQC, HMBC, and NOESY) spectroscopy. Their absolute configurations were confirmed on the basis of the absolute structure of 1, determined by a modified Mosher's method, and the absolute structure of 7, determined previously by X-ray crystallography.<sup>24</sup> The relative structure of 5 was further proven by X-ray diffraction analysis. The full <sup>1</sup>H and <sup>13</sup>C NMR data of 7 are reported herein for the first time. Cytotoxicity of metabolites 1-4, 6, and 7 against Hep G2 (human liver carcinoma), MCF-7, MDA-MB-231 (human breast carcinoma), and A-549 (human lung carcinoma) is also discussed.

## **Results and Discussion**

The EtOAc extract of the frozen animal was fractionated by silica gel column chromatography, and the eluted fractions were subjected to further separation and purification utilizing normal-phase HPLC to yield cembranolides  $1{\text -}7$  (see Experimental Section). All new metabolites were isolated as colorless oils (except 5) and showed positive optical rotations. Spectroscopic data revealed that all of these metabolites are cembranoids containing  $\alpha$ -methylene- $\gamma$ -lactones.

The HRFABMS  $(m/z, 393.2275 \text{ [M + H]}^+)$  of the most polar metabolite, crassocolide A (1), established the molecular formula C<sub>22</sub>H<sub>32</sub>O<sub>6</sub>, requiring seven degrees of unsaturation. The IR spectrum of 1 revealed the presence of hydroxy ( $\nu_{\text{max}}$  3440 cm<sup>-1</sup>),  $\alpha$ -methylene- $\gamma$ -lactone ( $\nu_{\rm max}$  1753 and 1653 cm<sup>-1</sup>), $^{2-17}$  and ester ( $\nu_{\rm max}$  1723 cm<sup>-1</sup>) moieties. The ion peaks at m/z 357 [M - 2 H<sub>2</sub>O + H]<sup>+</sup>, 297  $[M - 2 H_2O - AcOH + H]^+$  in the FABMS indicated the presence of two hydroxyls and one acetoxyl in 1. The latter was indicated from the  $^{1}H$  NMR data at  $\delta$  2.01 (3H, s) and the  $^{13}C$  NMR data at  $\delta_{\rm C}$  20.8 (CH<sub>3</sub>) and 169.5 (qC). The 22 carbon signals appearing in the <sup>13</sup>C NMR spectrum of **1** (Table 1) were identified by DEPT to belong to four methyls, six methylenes (including one olefinic carbon), six methines (including two vinylic CH), and six quaternary carbons. These data suggested 1 was a diterpenoid containing an acetoxy function. The NMR signals observed at  $\delta_{\rm C}$  170.0 (qC), 139.0 (qC), 122.6 (CH<sub>2</sub>), 80.9 (CH), and 36.1 (CH) and  $\delta_{\rm H}$  6.27, 5.70 (each, 1H, d, J = 2.5 Hz), 4.64 (1H, br s), and 3.31 (1H, m) revealed the presence of the  $\alpha$ -methylene- $\gamma$ -lactonic group as compared to those of similar metabolites.<sup>2-17</sup> Two trisubstituted double bonds were also designated from the NMR signals at  $\delta_{\rm C}$ 136.7 (qC), 130.4 (qC), 127.4 (CH), and 125.4 (CH) and at  $\delta_{\rm H}$ 5.29 (1H, dd, J = 7.0, 4.5 Hz) and 5.09 (1H, dd, J = 7.0, 5.0 Hz). The three 3H singlets appearing in the <sup>1</sup>H NMR at  $\delta$  1.74, 1.68, and 1.12 were assigned to two olefinic methyls and one methyl bound to a quaternary oxycarbon, respectively. The remaining one

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Table 1. <sup>13</sup>C NMR Data for Compounds 1-7

C#	$1^a$	$2^a$	$3^b$	$4^{a}$	<b>5</b> <sup>c</sup>	<b>6</b> <sup>c</sup>	$7^a$
1	36.1 (CH) <sup>d</sup>	37.2 (CH)	42.0 (CH)	41.7 (CH)	44.4 (CH)	44.1 (CH)	44.9 (CH)
2	36.6 (CH <sub>2</sub> )	36.4 (CH <sub>2</sub> )	35.6 (CH <sub>2</sub> )	36.1 (CH <sub>2</sub> )	28.4 (CH <sub>2</sub> )	26.8 (CH <sub>2</sub> )	32.6 (CH <sub>2</sub> )
3	74.9 (CH)	75.1 (CH)	74.9 (CH)	75.2 (CH)	119.6 (CH)	124.4 (CH)	120.1 (CH)
4	74.3 (qC)	75.5 (qC)	75.5 (qC)	74.2 (qC)	138.5 (qC)	136.3 (qC)	138.2 (qC)
5	37.5 (CH <sub>2</sub> )	37.9 (CH <sub>2</sub> )	38.1 (CH <sub>2</sub> )	37.6 (CH <sub>2</sub> )	39.1 (CH <sub>2</sub> )	79.1 (CH)	38.9 (CH <sub>2</sub> )
6	22.4 (CH <sub>2</sub> )	22.6 (CH <sub>2</sub> )	22.6 (CH <sub>2</sub> )	22.3 (CH <sub>2</sub> )	24.4 (CH <sub>2</sub> )	30.3 (CH <sub>2</sub> )	24.7 (CH <sub>2</sub> )
7	125.4 (CH)	123.6 (CH)	124.8 (CH)	125.6 (CH)	125.6 (CH)	120.4 (CH)	126.4 (CH)
8	136.7 (qC)	137.2 (qC)	137.0 (qC)	136.9 (qC)	133.3 (qC)	135.7 (qC)	133.1 (qC)
9	38.0 (CH <sub>2</sub> )	37.7 (CH <sub>2</sub> )	38.4 (CH <sub>2</sub> )	38.1 (CH <sub>2</sub> )	36.6 (CH <sub>2</sub> )	36.7 (CH <sub>2</sub> )	36.2 (CH <sub>2</sub> )
10	24.0 (CH <sub>2</sub> )	24.8 (CH <sub>2</sub> )	25.0 (CH <sub>2</sub> )	24.2 (CH <sub>2</sub> )	24.0 (CH <sub>2</sub> )	24.2 (CH <sub>2</sub> )	24.4 (CH <sub>2</sub> )
11	127.4 (CH)	128.1 (CH)	128.0 (CH)	128.2 (CH)	61.5 (CH)	61.5 (CH)	58.5 (CH)
12	130.4 (qC)	129.9 (qC)	131.0 (qC)	130.5 (qC)	60.5 (qC)	60.8 (qC)	57.4 (qC)
13	76.4 (CH)	77.5 (CH)	45.3 (CH <sub>2</sub> )	44.9 (CH <sub>2</sub> )	45.1 (CH <sub>2</sub> )	44.9 (CH <sub>2</sub> )	41.9 (CH <sub>2</sub> )
14	80.9 (CH)	81.3 (CH)	81.7 (CH)	80.9 (CH)	80.1 (CH)	80.1 (CH)	77.8 (CH)
15	139.0 (qC)	139.9 (qC)	139.7 (qC)	139.3 (qC)	138.0 (qC)	137.2 (qC)	139.2 (qC)
16	170.0 (qC)	169.7 (qC)	170.1 (qC)	170.3 (qC)	170.2 (qC)	170.1 (qC)	170.0 (qC)
17	122.6 (CH <sub>2</sub> )	121.2 (CH <sub>2</sub> )	122.9 (CH <sub>2</sub> )	123.1 (CH <sub>2</sub> )	121.9 (CH <sub>2</sub> )	122.5 (CH <sub>2</sub> )	121.6 (CH <sub>2</sub> )
18	25.6 (CH <sub>3</sub> )	24.9 (CH <sub>3</sub> )	24.8 (CH <sub>3</sub> )	25.8 (CH <sub>3</sub> )	14.8 (CH <sub>3</sub> )	11.4 (CH <sub>3</sub> )	15.4 (CH <sub>3</sub> )
19	17.1 (CH <sub>3</sub> )	16.9 (CH <sub>3</sub> )	16.2 (CH <sub>3</sub> )	16.8 (CH <sub>3</sub> )	14.9 (CH <sub>3</sub> )	15.1 (CH <sub>3</sub> )	15.6 (CH <sub>3</sub> )
20	15.2 (CH <sub>3</sub> )	14.6 (CH <sub>3</sub> )	17.3 (CH <sub>3</sub> )	17.5 (CH <sub>3</sub> )	17.5 (CH <sub>3</sub> )	17.6 (CH <sub>3</sub> )	19.1 (CH <sub>3</sub> )
Ac	20.8 (CH <sub>3</sub> )	20.7 (CH <sub>3</sub> )	21.3 (CH <sub>3</sub> )			21.6 (CH <sub>3</sub> )	
	169.5 (qC)	169.4 (qC)	170.7 (qC)			170.4 (qC)	
		21.2 (CH <sub>3</sub> )					
		170.9 (qC)					

"Spectra recorded at 125 MHz at 25 °C. The values are in ppm downfield from TMS. "Spectra recorded at 100 MHz at 25 °C. "Spectra recorded at 75 MHz in CDCl<sub>3</sub> at 25 °C. <sup>d</sup>Attached protons were determined by DEPT experiments.

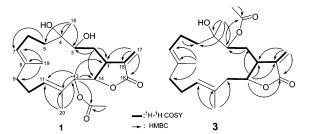
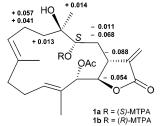


Figure 1. <sup>1</sup>H-<sup>1</sup>H COSY and HMBC correlations for 1 and 3.

degree of unsaturation was thus attributed to the 14-membered ring. Two oxymethines observed at  $\delta_{\rm C}$  76.4 (CH) and 74.9 (CH) and  $\delta_{\rm H}$ 5.40 (1H, s) and 3.52 (1H, d, J = 9.5 Hz) and an oxycarbon observed at  $\delta_{\rm C}$  74.3 (qC) indicated the presence of one acetoxy and two hydroxy substituents on the macrocyclic ring of 1. In the ¹H−¹H COSY spectrum, it was found that a ring-juncture methine proton ( $\delta$  3.31) exhibited allylic correlations with the exomethylene protons at C-17 ( $\delta_{\rm H}$  6.27 and 5.70, each d, J=2.5 Hz) and was assigned as H-1. Also, the <sup>1</sup>H-<sup>1</sup>H COSY spectral analysis established three partial structures of consecutive proton spin systems (Figure 1). Moreover, it was found that the downfield shifted oxymethine proton at  $\delta$  5.40 showed  ${}^{1}H/{}^{13}C$  long-range correlations in the HMBC spectrum with C-1 (δ 36.1, CH), C-14 ( $\delta$  80.9), the acetate carbonyl carbon ( $\delta$  169.5), and two olefinic carbons ( $\delta$  130.4, qC and 127.4, CH), implying the C-13 location of the acetoxy group and hence the presence of the C-11-C-12 olefin. Further analysis of the <sup>1</sup>H-<sup>1</sup>H COSY and HMBC correlations established the planar structure of 1, including the C-3 and C-4 positions of the remaining two hydroxy groups and the C-8 and C-9 location of the second olefinic moiety, as shown in Figure

In order to resolve the absolute configuration of 1, we determined the absolute configuration at C-3 using a modified Mosher's method.<sup>25,26</sup> The (S)- and (R)-2-methoxy-2-(trifluoromethyl)-2phenylacetic (MTPA) esters for 1 (1a and 1b, respectively) were prepared by using the corresponding (-)- and (+)-MTPA-chloride, respectively. The determination of  $\Delta \delta$  values  $(\delta_S - \delta_R)$  for the protons neighboring C-3 led to the assignment of the S configuration at C-3 in 1 (Figure 2). An NOE correlation (Figure 3) was observed between H-3 ( $\delta$  3.52, d, J = 9.5 Hz) and H<sub>3</sub>-18 ( $\delta$  1.12, s). Also, the <sup>1</sup>H NMR spectrum of 1 measured in C<sub>5</sub>D<sub>5</sub>N showed a downfield



**Figure 2.** <sup>1</sup>H NMR chemical shift differences  $\Delta \delta$  ( $\delta_S - \delta_R$ ) in ppm for the MTPA esters of **1**.

shift for H<sub>3</sub>-18 by the pyridine-induced deshielding effect ( $\Delta \delta$  =  $\delta \text{CDCl}_3 - \delta \text{C}_5 \text{D}_5 \text{N} = -0.32 \text{ ppm}$ ). This downfield shift value is equivalent only for a geminal deshielding resulting from 4-OH and not further deshielded by the vicinal 3-OH, revealing that 4-CH<sub>3</sub> should be anti to the 3-OH.27 By the above observations and NOE correlations between H-3 and H<sub>3</sub>-18, it was found that if H-3 is positioned on the  $\beta$ -face, then the 4-CH<sub>3</sub> should be *gauche* to H-3 and located on the  $\alpha$ -face (Figure 3). Furthermore, H-1 showed NOE correlation with H-3 but not with H-14. H-14 displayed NOE interactions with H-13, revealing the  $\beta$ -,  $\alpha$ -, and  $\alpha$ -orientations of H-1, H-13, and H-14, respectively. Thus, 1 possesses a trans-fused γ-lactone ring at C-1 and C-14. The NOE correlations observed for the two olefinic methyls at C-8 and C-12 with H<sub>2</sub>-6 and H<sub>2</sub>-10, respectively, reflected the E geometry of the two trisubstituted double bonds in the molecule. On the basis of the above findings and other detailed NOE correlations (Figure 3), the structure of crassocolide A (1) was fully elucidated and established as (1R,3S,4R,-13S,14R,7E,11E)-13-acetoxy-3,4-dihydroxycembra-7,11,15(17)trien-16,14-olide.

Crassocolide B (2) exhibited an ion peak at m/z 457.2204 [M + Na]+ in the HRESIMS, appropriate for a molecular formula of  $C_{24}H_{34}O_7$ , with one more unsaturation degree relative to that of 1. The IR spectrum also revealed the presence of an  $\alpha$ -methylene- $\gamma$ lactone ( $\nu_{max}$  1750 and 1653 cm<sup>-1</sup>). The presence of one hydroxyl and two acetoxyls was suggested from the IR absorptions ( $\nu_{max}$ 3447, 1734, and 1717 cm<sup>-1</sup>) and the ESIMS spectrum (m/z 315  $[M - 2 AcOH + H]^+$  and 297  $[M - 2 AcOH - H_2O + H]^+$ ). The NMR spectral data were found to be very similar to those of 1 (Tables 1 and 2), except for the presence of an additional acetoxyl ( $\delta_C$  21.2, CH<sub>3</sub> and 170.9, qC;  $\delta_H$  2.09, 3H, s) in 2, which replaces

Figure 3. Selected NOESY correlations of 1-4.

a hydroxyl in **1**. The acetoxy group was located at C-3 due to the downfield shift observed for H-3 ( $\delta$  4.78, dd, J = 6.0, 6.0 Hz) of **2** relative to that of **1** ( $\delta$  3.52, d, J = 9.5 Hz). Moreover, acetylation of **1** with acetic anhydride in pyridine yielded a product that showed identical  $^1$ H NMR data to that of **2**, including the same splitting pattern and J value of H-3, and thus confirmed that **2** is the O-acetyl derivative of **1**. On the basis of the above findings and the detailed analysis of the key NOESY correlations (Figure 3), the structure of crassocolide B (**2**) was established as (1R,3S,4R,13S,14R,7E,-11E)-3,13-diacetoxy-4-hydroxycembra-7,11,15(17)-trien-16,14-olide.

Crassocolide C (3) was found to possess the molecular formula  $C_{22}H_{32}O_5$  from the HRFABMS (m/z 377.2325 [M + H]<sup>+</sup>). The presence of the  $\alpha$ -methylene- $\gamma$ -lactone, acetoxy, and hydroxy functionalities was indicated by IR ( $\nu_{\rm max}$  1748 and 1660, 1715, and 3422 cm<sup>-1</sup>) and FABMS, which showed ion peaks at m/z 359 [M - H<sub>2</sub>O + H]<sup>+</sup> and 317 [M - AcOH + H]<sup>+</sup>. The <sup>13</sup>C NMR data of 3 were found to be very similar to those of 2 (Tables 1), except for the replacement of the three carbon signals of an acetoxymethine moiety in 2 by the signal of a methylene carbon in 3 ( $\delta$  45.3, CH<sub>2</sub>, C-13). The similar chemical shifts of C-2 and C-3 in both 2 and 3 positioned the acetate group at C-3 in 3. Furthermore, it was found that the splitting pattern and J value of the acetoxymethine proton H-3 ( $\delta$  4.84, dd, J = 6.2, 6.2 Hz) of **3** were similar to those of **2** ( $\delta$  4.78, dd, J = 6.0, 6.0 Hz), implying the very similar partial structure around C-3 in both 2 and 3. The overall planar structure of 3 was established by analyzing the <sup>1</sup>H-<sup>1</sup>H COSY and HMBC correlations (Figure 1). The above observations together with the very similar NOE correlations for both 2 and 3 revealed that they have the same configurations at C-1, C-3, C-4, and C-14, respectively. The structure of crassocolide C (3) was thus established as (1R,3S,4R,14S,7E,11E)-3-acetoxy-4-hydroxycembra-7,11,15(17)trien-16,14-olide.

Crassocolide D (4) possessed the molecular formula  $C_{20}H_{30}O_4$  as revealed from its HRFABMS (m/z 335.2222 [M + H]<sup>+</sup>). The IR absorptions at  $\nu_{max}$  1750, 1653, and 3422 cm<sup>-1</sup> and the ion peaks at m/z 317 [M - H<sub>2</sub>O + H]<sup>+</sup> and 299 [M - 2 H<sub>2</sub>O + H]<sup>+</sup> indicated the presence of the  $\alpha$ -methylene- $\gamma$ -lactone and two hydroxyl groups.

Comparison of the <sup>1</sup>H and <sup>13</sup>C NMR spectral data of **4** with those of **3** revealed **4** as the deacetyl derivative of **3**. It was found that H-3 of **4** resonated at higher field ( $\delta$  3.53, dd, J = 9.5, 7.5 Hz) relative to that of **3** ( $\delta$  4.84, dd, J = 6.2, 6.2 Hz). Thus, the acetoxy group at C-3 of **3** was replaced by a hydroxy group in **4**. It was observed that the NOE correlations of **4** (Figure 3) are very similar to those of **1**. Thus, the structure of crassocolide D (**4**) was deduced as (1R,3S,4R,14S,7E,11E)-3,4-dihydroxycembra-7,11,15(17)-trien-16,14-olide.

The least polar metabolite, crassocolide E (5), was isolated as colorless prisms. Its HRFABMS (m/z 317.2117,  $[M + H]^+$ ) and NMR data (Tables 1 and 2) established the molecular formula C<sub>20</sub>H<sub>28</sub>O<sub>3</sub>, implying seven degrees of unsaturation. The IR absorptions of 5 at  $\nu_{\rm max}$  1765 and 1662 cm<sup>-1</sup> also revealed the presence of an  $\alpha$ -methylene- $\gamma$ -lactone moiety. This was further indicated from the <sup>1</sup>H NMR signals at  $\delta$  6.24 and 5.59 (each 1H, d, J = 3.0Hz) and  ${}^{13}$ C NMR signals at  $\delta$  170.2 (qC, C-16), 138.0 (qC, C-15), 121.9 (CH<sub>2</sub>, C-17), 80.1 (CH, C-14), and 44.4 (CH, C-1). Moreover, the  ${}^{13}\text{C}$  NMR signals at  $\delta$  138.5 (qC), 133.3 (qC), 125.6 (CH), 119.6 (CH), 61.5 (CH), and 60.5 (qC) assigned two trisubstituted double bonds and one trisubstituted epoxide in the molecule, respectively. The 3H singlets appearing in the <sup>1</sup>H NMR spectrum at  $\delta$  1.66, 1.51, and 1.41 represented the methyl substituents at the two double bonds and the epoxide, respectively. By interpretation of <sup>1</sup>H-<sup>1</sup>H COSY correlations, it was possible to establish three partial structures of consecutive proton systems extending from the olefinic proton H-3 to H<sub>2</sub>-13 through H-1 and H-14, from H<sub>2</sub>-5 to the olefinic H-7, and from H<sub>2</sub>-9 to H-11; the connectivities of these partial structures were further established by the HMBC correlations (Figure 4). The long-range <sup>1</sup>H/<sup>13</sup>C correlations observed from the methyl protons at  $\delta$  1.41 (H<sub>3</sub>-20) to C-11 ( $\delta$  61.5, CH), C-12 (60.5, qC), and C-13 ( $\delta$  45.1, CH<sub>2</sub>) indicated the C-11-C-12 position of the epoxide. On the basis of the above observations, and additional 2D NMR (<sup>1</sup>H-<sup>1</sup>H COSY, HMQC, and HMBC) experiments, the planar structure of 5 was established as illustrated in Figure 4. On comparison of chemical shifts and splitting patterns of the protons at C-1, C-3, C-7, C-11, C-14, and C-17-C-20 in 5 with the corresponding protons in lobophytolide (7), of which the structure was established by X-ray diffraction analysis, 23,24 it was found that the difference between these compounds was the configuration at the epoxide moiety. This was due to the variation in chemical shifts and splitting patterns for H-11 ( $\delta$  2.61, 1H, dd, J = 7.5, 1.8 Hz) and  $H_3$ -20 ( $\delta$  1.41, 3H, s) in **5** relative to those in **7** ( $\delta$  2.81, 1H, dd, J = 6.0, 6.0 Hz and 1.28, 3H, s, respectively). Finally, the structure of 5 was unambiguously established by single-crystal X-ray diffraction analysis (Figure 5) as (1R,11R,12R,14S,3E,7E)-11,12-epoxycembra-3,7,15(17)-trien-16,14-olide.

The new metabolite crassocolide F (6) was found to have the molecular formula C<sub>22</sub>H<sub>30</sub>O<sub>5</sub> from its HRESIMS (m/z 397.1989,  $[M + Na]^+$ ) and NMR data (Table 1 and 2), implying eight degrees of unsaturation. Similar to 1-5, the IR spectrum of 6 indicated the presence of an  $\alpha$ -methylene- $\gamma$ -lactone group ( $\nu_{max}$  1767 and 1667 cm<sup>-1</sup>), and this was further supported from NMR shifts at  $\delta_{\rm H}$  6.40 and 5.62 (each 1H, d, J = 3.0 Hz);  $\delta_{\rm C}$  170.1, 137.2, 122.5, 80.1, and 44.1. An acetoxy group was also revealed from the ESIMS  $(m/z \ 315 \ [M - AcOH + H]^{+})$  and IR  $(\nu_{max} \ 1734 \ cm^{-1})$  spectra and was supported by the <sup>1</sup>H NMR signals at  $\delta$  2.03 (3H, s). The NMR data of 6 showed signals of high similarity to those of 5 except for the replacement of a methylene group ( $\delta_{\rm H}$  2.27, 2H, m;  $\delta_{\rm C}$  39.1) in **5** with an acetoxymethine signal ( $\delta_{\rm H}$  5.24, 1H, dd, J=10.5, 4.8 Hz and 2.03, 3H, s;  $\delta_{\rm C}$  170.4, qC,  $\delta$  79.1, CH, and 21.6, CH<sub>3</sub>) in **6**. The HMBC correlations observed from the methyl protons at  $\delta$  1.43 (H<sub>3</sub>-20) to C-11 ( $\delta$  61.5, CH), C-12 (60.8, qC), and C-13 ( $\delta$  44.9, CH<sub>2</sub>) and from the olefinic methyl protons at  $\delta$ 1.54 to C-3 ( $\delta$  124.4, CH), C-4 ( $\delta$  136.3, qC), and C-5 ( $\delta$  79.1, CH) revealed that the epoxide and the acetoxy groups should be positioned at C-11-C-12 and C-5, respectively. On the basis of

Table 2. <sup>1</sup>H NMR Data for Crassocolides 1-7<sup>d</sup>

H#	$1^a$	$2^a$	$3^{b}$	<b>4</b> a	$5^c$	<b>6</b> <sup>c</sup>	$7^a$
1	3.31 m (w <sub>1/2</sub>	3.32 m (w <sub>1/2</sub>	3.10 m (w <sub>1/2</sub>	2.99 m (w <sub>1/2</sub>	2.77 m (w <sub>1/2</sub>	2.80 m (w <sub>1/2</sub>	2.64 m (w <sub>1/2</sub>
	$(12.0)^{c}$	11.5)	11.0)	11.0)	11.5)	12.5)	12.0)
2	1.54 m; 1.77 m	1.83 m; 2.01 m	1.88 m; 1.94 m	1.58 m; 1.74 m	2.30 2H, m	2.45 m; 2.49 m	2.27 2H, m
3	3.52 d (9.5)	4.78 dd (6.0,	4.84 dd (6.2, 6.2)	3.53 dd (9.5, 7.5)	4.98 dd (6.6,	5.35 dd (6.9,	5.06 dd (7.0,
		6.0)			5.0)	$(6.9)^c$	7.0)
5	1.71 2H, m	1.74 m; 1.79 m	1.72 2H, m	1.73 2H, m	2.27 2H, m	5.24 dd (10.5, 4.8)	2.19 2H, m
6	2.06 m; 2.42 m	2.05 m; 2.27 m	2.09 m; 2.18 m	2.04 m; 2.46 m	2.10 m; 2.38 m	2.34, m; 2.55 m	2.21 m; 2.29 m
7	5.09 dd (7.0,	5.23 dd (7.0,	5.25 dd (7.0,	5.10 dd (8.0,	4.98 br dd (6.6,	4.90 d (9.9)	5.01 br dd (6.0,
	5.0)	7.0)	7.0)	4.5)	5.0)		6.0)
9	2.16 m; 2.26 m	2.12 m; 2.21 m	2.18 2H, m	2.15 m; 2.21 m	2.06 m; 2.22 m	2.05 m; 2.32 m	2.09 m; 2.22 m
10	2.18; 2.39 m	2.20; 2.41 m	2.22 m; 2.34 m	2.14; 2.31 m	1.31 m; 2.06 m	1.34 m; 2.12 m	1.53 m; 1.78 m
11	5.29 d (7.0,	5.31 dd (7.0,	5.06 br dd (6.0,	5.09 dd (6.5,	2.61 dd (7.5,	2.53 m	2.81 dd (6.0,
	4.5)	7.0)	6.0)	6.5)	1.8)		6.0)
13	5.40 s	5.38 s	2.08 m; 2.48 m	2.12 dd (14.0, 10.0)	1.31 m; 2.26 m	1.32 m; 2.29 m	2.04 m; 2.09 m
				2.64 br d (14.0)			
14	4.64 br s	4.63 br s	4.44 ddd (10.0,	4.52 ddd (10.5,	4.26 ddd (9.0,	4.12 ddd (9.9,	3.96 dd (11.0,
			5.0, 5.0)	4.0, 3.0)	6.6, 1.5)	6.0, 1.5)	5.0)
17	5.70 d (2.5)	5.67 d (2.0)	5.68 d (1.8)	5.68 d (2.5)	5.59 d (3.0)	5.62 d (3.0)	5.59 d (2.0)
	6.27 d (2.5)	6.23 d (2.0)	6.26 d (1.8)	6.27 d (2.5)	6.24 d (3.0)	6.40 d (3.0)	6.24 d (2.0)
18	1.12 3H, s	1.19 3H, s	1.19 3H, s	1.13 3H, s	1.51 3H, s	1.54 3H, s	1.58 3H, s
19	1.68 3H, s	1.68 3H, s	1.62 3H, s	1.66 3H, s	1.66 3H, s	1.69 3H, s	1.66 3H, s
20	1.74 3H, s	1.77 3H, s	1.74 3H, s	1.70 3H, s	1.41 3H, s	1.43 3H, s	1.28 3H, s
Ac	2.01 3H, s	1.99 3H, s	2.07 3H, s			2.03 3H, s	
		2.09 3H, s					

"Spectra recorded at 500 MHz at 25 °C. "Spectra recorded at 400 MHz at 25 °C. "Spectra recorded at 300 MHz in CDCl3 at 25 °C. "The J values are in Hz in parentheses.

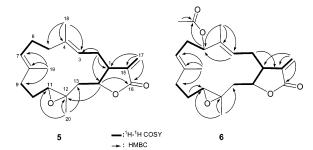


Figure 4. <sup>1</sup>H-<sup>1</sup>H COSY and HMBC correlations for 5 and 6.

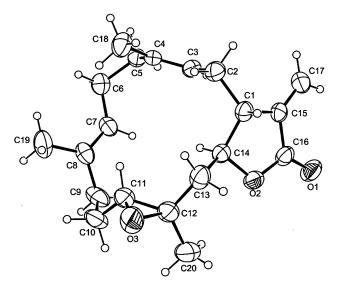


Figure 5. Molecular structure of 5 based on X-ray analysis.

the above observations and additional 2D NMR (<sup>1</sup>H-<sup>1</sup>H COSY, HMQC, and HMBC) experiments, the planar structure of 6 was established as shown in Figure 4.

The configurations of the five chiral centers at C-1, C-5, C-11, C-12, and C-14 in 6 were determined on the basis of NOE correlations observed from a NOESY experiment (Figure 6). It was found that H-11 ( $\delta$  2.53, m), but not H<sub>3</sub>-20, showed NOE interaction with the α-oriented H-14 (as deduced from the absolute configu-

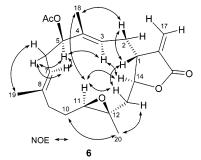


Figure 6. Observed NOESY correlations of 6.

ration of C-14 in the biogenetically related metabolite 1), implying the downward orientation of H-11. The methyl protons H<sub>3</sub>-20 showed NOE correlation with H<sub>2</sub>-10, but not with H-11 and thus assigned *trans* geometry to the epoxide ring. The  $\beta$ -orientation of the epoxide was established, since the NMR data of C-11-C-14 and C-20 and their attached protons in 6 were quite similar to those in 5. Moreover, H-14 exhibited an NOE interaction with H-3 ( $\delta$ 5.35, dd, J = 6.9, 6.9 Hz) and H-3 showed an NOE interaction with H-5 ( $\delta_{\rm H}$  5.24, 1H, dd, J=10.5, 4.8 Hz), revealing the  $\alpha$ -orientation of H-5, and thus the  $\beta$ -orientation of the acetoxy group at C-5 as shown in Figure 5. The two upfield shifted olefinic methyls that exhibited carbon signals below 20 ppm, together with the NOE interactions observed between H<sub>3</sub>-19 and H-6 and between H<sub>3</sub>-18 and H-2, confirmed the E geometry of the two trisubstituted double bonds. Further interpretation of NOESY data (Figure 6) again indicated the trans-fused lactone ring and hence established the stereochemistry of 6. On the basis of the above results, the structure of crassocolide F was elucidated as (1R,5R,11R,-12R14S,7E,3E)-5-acetoxy-11,12-epoxycembra-3,7,15(17)-trien-16,-

Compound 7 [(1R,11S,12S,14S,7E,3E)-11,12-epoxycembra-3,7,-15(17)-trien-16,14-olide] was also isolated from S. crassocaule and found to be identical to the previously reported metabolite lobophytolide (7), isolated from the soft coral *Lobophytum cristagalli*, <sup>26</sup> by comparison of the physical (mp and  $[\alpha]_D$ ) and spectroscopic (IR, MS, <sup>1</sup>H NMR) data. Moreover, by 2D NMR analysis, the full assignment of NMR data for 7 is reported here for the first time (Tables 1 and 2).

**Table 3.** Cytotoxicity (IC<sub>50</sub>  $\mu$ g/mL) of Crassocolides 1–4, 6, and 7

	Hep G2	MCF-7	MDA-MB-231	A549
1	3.1	8.9	8.6	11.9
2	13.1	10.3	>30	12.1
3	$> 30^{a}$	>30	>30	>30
4	>30	15.3	>30	12.5
6	2.1	7.4	8.8	3.2
7	6.3	2.3	2.0	2.1
doxorubicin	0.2	0.3	0.1	0.2

<sup>&</sup>lt;sup>a</sup> Compound is considered inactive when  $IC_{50} > 30 \mu g/mL$ .

The cytotoxicity of the diterpenoids 1-4, 6, and 7 against HepG2, MCF7, MDA-MB-231, and A-549 cancer cells was studied. The results (Table 3) showed that the 13-acetoxy-3,4-dihydroxycembranolide (1) was cytotoxic against the four cancer cell lines, being significant (IC<sub>50</sub> 3.1  $\mu$ g/mL) to moderate (IC<sub>50</sub> 8.6–11.9  $\mu$ g/ mL) against Hep G2, and MCF-7, MDA-MB-231, and A549, respectively. The acetylation of 1 at the 4-OH position resulted in the formation of 2 and reduced cytotoxicity. The loss of the 13acetoxy group from 2 to yield 3 was associated with a complete loss of cytotoxic activity against the four cell lines. Compound 4, which was not oxidized at C-13, but possessed two hydroxy groups at C-3 and C-4 like that of 1, retained partial cytotoxicity and exhibited moderate activity against two cell lines (MCF-7 and A549). The 5-O-acetylated epoxycembranolide 6 was strongly cytotoxic against Hep G2 and A549 but showed only moderate activity against MCF-7 and MDA-MB-231 cancer cell lines. While the known epoxycembranolide (7) showed only moderate cytotoxicity against Hep G2 cells (IC50 6.3 µg/mL), it was found to exhibit significant cytotoxicity (IC<sub>50</sub>  $\approx 2.0 \mu \text{g/mL}$ ) against other cancer cell lines.

## **Experimental Section**

General Experimental Procedures. Melting points were determined using a Fisher-Johns melting point apparatus. Optical rotations were measured on a Jasco DIP-1000 digital polarimeter. IR spectra were recorded on a Hitachi I-2001 infrared spectrophotometer. FABMS were obtained with a VG Quattro GC/MS spectrometer. HRFABMS spectra were recorded on a JEOL-SX/SX 102A mass spectrometer. The NMR spectra were recorded on a Bruker AMX-400 FT-NMR at 400 MHz for  $^1\text{H}$  and 100 MHz for  $^1\text{H}$  and 125 MHz for  $^{13}\text{C}$ , respectively, in CDCl<sub>3</sub> using TMS as internal standard. The crystallographic data were collected on a Rigaku AFC7S diffractometer using graphite-monochromated Mo Kα radiation. Structure analysis was made by using the SHELXTL PLUS package. Si gel 60 (Merck, 230–400 mesh) was used for CC. Precoated Si gel plates (Merck, Kieselgel 60 F254, 0.2 mm) were used for analytical TLC analyses.

**Animal Material.** *S. crassocaule* (Alcyoniidea) (1.05 kg, wet wt) was collected by hand via scuba at the coast of Kenting, in December 2001 and July 2002, at a depth of 10 to 15 m, and stored in a freezer until extraction. A voucher sample was deposited at the Department of Marine Resources, Sun Yat-sen University.

**Extraction and Separation.** The frozen bodies of *S. crassocaule* (1.05 kg, wet wt) were minced and exhaustively extracted with EtOAc (1 L × 5). The solvent-free EtOAc extract (15.0 g) was subjected to Si gel CC and eluted with EtOAc in n-hexane (0-100%, gradient) to yield 20 fractions. Fraction 10 eluted with EtOAc-n-hexane (1:10) and was purified by normal-phase HPLC using acetone-n-hexane (1: 20) to give 5 (27 mg). Fraction 12 eluted with EtOAc-n-hexane (1:5) and was further chromatographed by normal-phase HPLC using acetone-n-hexane (1:13) to give 7 (27.8 mg). Fraction 14 eluted with EtOAc-n-hexane (2:5) and was further separted by normal-phase HPLC using acetone—n-hexane—CH<sub>2</sub>Cl<sub>2</sub> (1:15:2) to yield **2** (3.8 mg) and 6 (2.2 mg). Fraction 15 eluted with n-hexane-EtOAc (1:1) and was further purified by normal-phase HPLC using acetone-n-hexane (1:6) to afford 3 (5.5 mg). The next two successive fractions eluted with EtOAc-n-hexane (1:1 to 2:1) and were separately rechromatographed by normal-phase HPLC using acetone—*n*-hexane (1:4) to afford 4 (56.5 mg) and 1 (47.3 mg), respectively.

**Crassocolide A (1):** colorless oil;  $[α]^{25}_D$  +7.0 (c 2.7, CHCl<sub>3</sub>); IR (neat)  $ν_{max}$  3440, 2958, 2928, 2878 1753, 1723,1653, 1437, 1373, 1273, and 1232 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data (CDCl<sub>3</sub>), see Tables 2 and 1, respectively; <sup>1</sup>H NMR (C<sub>5</sub>D<sub>5</sub>N) δ 6.29 (1H, s, H-17), 5.94 (1H, s, H-17), 5.63 (1H, s, H-13), 5.62 (1H, m, H-11), 5.49 (1H, m, H-7), 5.49 (1H, br s, H-14), 3.98 (1H, br d, J=9.3 Hz, H-3), 3.47 (1H, m, H-1), 2.66 (1H, m, H-6), 2.38 (1H, br d, J=14.0 Hz, H-2), 2.28 (2H, m, H-9, H-10), 2.18 (2H, m, H-9, 10), 2.13 (1H, m, H-5), 197 (3H, s, Ac), 1.93 (1H, m, H-2), 1.89 (3H, s, H<sub>3</sub>-20), 1.88 (2H, m, H-5, H-6), 1.63 (3H, s, H<sub>3</sub>-19), 1.44 (3H, s, H<sub>3</sub>-18); FABMS m/z 393 [0.6, (M + H)<sup>+</sup>], 357 [0.2, (M - 2 H<sub>2</sub>O + H)<sup>+</sup>], 297 [0.6, (M - 2 H<sub>2</sub>O - AcOH + H)<sup>+</sup>], 273 (1.9), 257 (2.0), 241 (1.6), and 229 (3.9); HRFABMS m/z 393.2275 (calcd for C<sub>22</sub>H<sub>33</sub>O<sub>6</sub>, 393.2277).

**Crassocolide B (2):** colorless oil;  $[\alpha]^{25}_D + 31.6$  (*c* 1.5, CHCl<sub>3</sub>); IR (neat)  $\nu_{\text{max}}$  3447, 2956, 2932, 2872 1750, 1734, 1717, 1653, 1456, 1397, 1260, and 1238 cm<sup>-1</sup>;  $^{1}\text{H}$  and  $^{13}\text{C}$  NMR data (CDCl<sub>3</sub>), see Tables 2 and 1, respectively; ESIMS m/z 457 [100, (M + Na)<sup>+</sup>], 375 [67, (M - AcOH + H)<sup>+</sup>], 315 [41, (M - 2 AcOH + H)<sup>+</sup>] and 297 [57, (M - 2 AcOH - H<sub>2</sub>O + H)<sup>+</sup>]; HRESIMS m/z 457.2204 (calcd for C<sub>24</sub>H<sub>34</sub>O<sub>7</sub>-Na, 457.2202).

**Crassocolide C (3):** colorless oil;  $[\alpha]^{25}_D + 19.1$  (*c* 1.1, CHCl<sub>3</sub>); IR (neat)  $\nu_{\text{max}}$  3422, 2958, 2932, 2880, 1748, 1715, 1660, 1456, 1375, and 1242 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data (CDCl<sub>3</sub>), see Tables 2 and 1, respectively; FABMS m/z 399 [0.2, (M + Na)<sup>+</sup>], 377 [0.4, (M + H)<sup>+</sup>], 359 [0.4, (M - H<sub>2</sub>O + H)<sup>+</sup>], 317 [0.9, (M - AcOH + H)<sup>+</sup>], 299 [0.5, (M - H<sub>2</sub>O - AcOH + H)<sup>+</sup>], 267 (0.8), and 253 (0.7); HRFABMS m/z 377.2325 (calcd for C<sub>22</sub>H<sub>33</sub>O<sub>5</sub>, 377.2328).

**Crassocolide D (4):** colorless oil;  $[\alpha]^{25}_{\rm D}$  +21.9 (*c* 1.3, CHCl<sub>3</sub>); IR (neat)  $\nu_{\rm max}$  3422, 2958, 2932, 2880, 1750, 1653, 1435, 1387, and 1269 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data (CDCl<sub>3</sub>), see Tables 2 and 1, respectively; FABMS m/z 357 [2.1, (M + Na)<sup>+</sup>], 335 [1.7, (M + H)<sup>+</sup>], 317 [19.6, (M - H<sub>2</sub>O + H)<sup>+</sup>], 299 [1.2, (M - 2 H<sub>2</sub>O + H)<sup>+</sup>]; HRFABMS m/z 335.2222 (calcd for C<sub>20</sub>H<sub>31</sub>O<sub>4</sub>, 335.2222).

**Crassocolide E (5):** colorless needles (EtOAc) mp 95–96 °C;  $[\alpha]^{25}$ D +108.9 (c 1.0, CHCl<sub>3</sub>); IR (neat)  $\nu_{\rm max}$  2960, 2932, 2872, 1765, 1662, 1437, 1386, 1267, and 1240 cm<sup>-1</sup>;  $^{1}$ H and  $^{13}$ C NMR data (CDCl<sub>3</sub>), see Tables 2 and 1, respectively; FABMS m/z 317 [6.0, (M + H)<sup>+</sup>]; HRFABMS m/z 317.2117 (calcd for C<sub>20</sub>H<sub>29</sub>O<sub>3</sub>, 317.2117).

**Crassocolide F (6):** colorless oil;  $[\alpha]^{25}_{\rm D}$  +138.3 (*c* 0.6, CHCl<sub>3</sub>); IR (neat)  $\nu_{\rm max}$  2956, 2926, 2870, 1767, 1734, 1667, 1437, 1372, and 1238 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data (CDCl<sub>3</sub>), see Tables 3 and 1, respectively; ESIMS m/z 397 [100, (M + Na)<sup>+</sup>], 381 [31, (M - O + Na)<sup>+</sup>], 315 [37, (M - AcOH + H)<sup>+</sup>]; HRESIMS m/z 397.1989 (calcd for C<sub>22</sub>H<sub>30</sub>O<sub>5</sub>-Na, 397.1991).

**Lobophytolide** (7): colorless needles; mp 133–134 °C (lit.,  $^{26}$  137–138 °C);  $[\alpha]^{25}_{\rm D}$  +23.1 (c 1.0, CHCl<sub>3</sub>) (lit.,  $^{26}$  c 0.4, +7.0); IR (neat)  $\nu_{\rm max}$  2928, 2851, 1752, 1655, 1543, 1385, 1338, and 1273 cm<sup>-1</sup>;  $^{1}$ H and  $^{13}$ C NMR data (CDCl<sub>3</sub>), see Tables 2 and 1, respectively; FABMS m/z 339 [16.9, (M + Na)+], 317 [12.6, (M + H)+].

**Preparation of (S)- and (R)-MTPA Esters of 1.** To a solution of 1 (4.7 mg, 12  $\mu$ M) in pyridine (50  $\mu$ L) was added (—)-MTPA chloride (5  $\mu$ L, 26  $\mu$ M), and the solution was allowed to stand overnight at room temperature. The reaction was stopped by the addition of 1.0 mL of water, followed by extraction with  $CH_2Cl_2$  (1.0 mL  $\times$  3). The CH<sub>2</sub>Cl<sub>2</sub>-soluble layers were combined, dried over anhydrous MgSO<sub>4</sub>, and evaporated. The residue was subjected to a short Si column using EOAC-n-hexane (2:5) to yield the (S)-MTPA ester 1a (1.1 mg, 18%) and an unreacted portion of 1 (3.0 mg). The same procedure was applied to obtain the (R)-MTPA ester 1b from the reaction of (+)-MTPA chloride with 1 in pyridine. <sup>1</sup>H NMR (CDCl<sub>3</sub>) of 1a:  $\delta$  7.517 (2H, m, Ph), 7.436 (3H, m, Ph), 6.213 (1H, d, J = 2.0 Hz, H-17), 5.677 (1H, d, J = 2.0 Hz, H-17), 5.357 (1H, s, H-13), 5.280 (1H, m, H-7), 5.231 (1H, m, H-11), 4.781 (1H, m, H-3), 4.698 (1H, s, H-14), 3.473 (3H, s, OMe), 3.377 (1H, br d, J = 9.3 Hz, H-1), 2.174 (1H, m, H-6), 1.970 (3H, s, OAc), 1.945 (1H, m, H-2), 1.812 (3H, H<sub>3</sub>-20), 1.794 (1H, m, H-6), 1.704 (1H, m, H-2), 1.644 (3H, H<sub>3</sub>-19), 1.614 (2H, m, H<sub>2</sub>-5), 1.162 (3H, H<sub>3</sub>-18). <sup>1</sup>H NMR (CDCl<sub>3</sub>) of **1b**:  $\delta$  7.517 (2H, m, Ph), 7.435 (3H, m, Ph), 6.207 (1H, d, J = 2.0 Hz, H-17), 5.683 (1H, d, J= 2.0 Hz, H-17), 5.346 (1H, s, H-13), 5.327 (1H, m, H-7), 5.254 (1H, m, H-11), 4.737 (1H, m, H-3), 4.652 (1H, s, H-14), 3.530 (3H, s, OMe), 3.465 (1H, br d, J = 9.9 Hz, H-1), 2.133 (1H, m, H-6), 2.013 (1H, m, H-2), 1.961 (3H, s, OAc), 1.857 (3H, H<sub>3</sub>-20), 1.737 (1H, m, H-6), 1.715 (1H, m, H-2), 1.681 (3H, H<sub>3</sub>-19), 1.601 (2H, m, H<sub>2</sub>-5), 1.148 (3H, H<sub>3</sub>-

Crystallographic Data and X-ray Structure Analysis of Crassocolide E (5). <sup>28</sup> A suitable colorless crystal (0.65 × 0.5 × 0.4 mm³) of 5 was grown by slow evaporation of the EtOAc solution. Crystal data: C<sub>20</sub>H<sub>28</sub>O<sub>3</sub>, monoclinic,  $M_r$  = 316.42 g/mol; a = 6.2085(12) Å, b = 9.7196(19) Å, c = 15.606(3) Å, V = 937.5(3) Å, ³ space group P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>, Z = 4,  $D_{\rm calc}$  = 1.121 g/cm³,  $\lambda$  = 0.71073 Å,  $\mu$ (Mo K $\alpha$ ) = 0.074 mm<sup>-1</sup>, F(000) = 344, T = 298(2) K. A total of 3691 reflections were collected in the range 2.47° <  $\theta$  < 26.01°, of which only 1961 unique reflections with I > 2 $\sigma$ (I) were used for the analysis. The structure was solved using direct methods and refined by full-matrix least-squares on F2 values. The non-hydrogen atoms were refined anisotropically. All hydrogen atoms were fixed at calculated positions. The final indices were  $R_1$  = 0.0412, wR<sub>2</sub> = 0.1199 with goodness-of-fit = 1.026.

**Cytotoxicity Testing.** Cell lines were purchased from the American Type Culture Collection (ATCC). Cytotoxicity assays were performed using the MTT [3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide] colorimetric method.<sup>29,30</sup>

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**Note Added after ASAP Publication:** The version posted on October 19, 2006 had a misspelling in the title. The correct spelling of Crassocolides appears in the version posted on November 8, 2006.

**Supporting Information Available:** Description of X-ray crystal structure data of compound **5**. This material is available free of charge via the Internet at http://pubs.acs.org.

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